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Manganese(II) octauranium(IV) heptadecasulfide

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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{Mn-S}) = 0.002$ Å; R factor = 0.026; wR factor = 0.068; data-to-parameter ratio = 22.0.

Single crystals of manganese(II) octauranium(IV) heptadecasulfide, $\text{MnU}_8\text{S}_{17}$, were grown from the reaction of the elements in a RbCl flux. $\text{MnU}_8\text{S}_{17}$ crystallizes in the space group $C2/m$ in the $\text{CrU}_8\text{S}_{17}$ structure type. The asymmetric unit is composed of the following atoms with site symmetries shown: U1 (1), U2 (m), U3 (m), Mn1 ($2/m$), S1 (1), S2 (1); S3 (m), S4 (m), S5 (m), S6 (m) and S7 ($2/m$). The three U^{IV} atoms are each coordinated by eight S atoms in a bicapped trigonal-prismatic arrangement. The Mn^{II} atom is coordinated by six S atoms in a distorted octahedral arrangement.

Related literature

$\text{MnU}_8\text{S}_{17}$ was previously determined to be isostructural with $\text{CrU}_8\text{S}_{17}$ (Noël *et al.*, 1975) from powder diffraction data (Noël, 1973). Single-crystal refinements have been carried out for the Cr, Fe (Kohlmann *et al.*, 1997), and Sc (Vovan & Rodier, 1979) analogues. Magnetic data for these compounds are available (Noël & Troc, 1979). The Mn–S distances are consistent with those expected for low-spin Mn^{II} (Shannon, 1976). For synthetic details, see: Bugaris & Ibers (2008); Haneveld & Jellinek (1969). For computational details, see: Gelato & Parthé (1987).

Experimental

Crystal data

$\text{MnU}_8\text{S}_{17}$
 $M_r = 2504.20$
 Monoclinic, $C2/m$

$a = 13.3549$ (6) Å
 $b = 8.3893$ (4) Å
 $c = 10.4927$ (5) Å

$\beta = 101.658$ (2)°
 $V = 1151.33$ (9) Å³
 $Z = 2$
 Mo $K\alpha$ radiation

$\mu = 58.10$ mm⁻¹
 $T = 100$ K
 $0.15 \times 0.09 \times 0.08$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: numerical face indexed (SADABS; Sheldrick, 2008a)
 $T_{\text{min}} = 0.043$, $T_{\text{max}} = 0.085$

8422 measured reflections
 1609 independent reflections
 1578 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.026$
 $wR(F^2) = 0.068$
 $S = 2.10$
 1609 reflections

73 parameters
 $\Delta\rho_{\text{max}} = 3.90$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.94$ e Å⁻³

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008b); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008b); molecular graphics: CrystalMaker (Palmer, 2009); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2512).

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supplementary materials

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Manganese(II) octauranium(IV) heptadecasulfide

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Comment

Single crystals of $\text{MnU}_8\text{S}_{17}$ were obtained in an attempt to synthesize a quaternary arsenic-containing compound (see Experimental).

$\text{MnU}_8\text{S}_{17}$ crystallizes in the $\text{CrU}_8\text{S}_{17}$ structure type (Noël *et al.*, 1975). Structure determinations based on single crystal data for isostructural $\text{ScU}_8\text{S}_{17}$ (Vovan & Rodier, 1979) and $\text{FeU}_8\text{S}_{17}$ (Kohlmann *et al.*, 1997) have also been reported, and $\text{MnU}_8\text{S}_{17}$ was also determined to be isostructural from powder diffraction experiments (Noël, 1973).

The structure comprises three independent U atoms, each coordinated by eight S atoms in a bicapped trigonal prismatic arrangement, and one independent Mn atom that is coordinated by six S atoms in a distorted octahedral arrangement (Figs. 1,2). There are no S—S bonds in the structure, so formal oxidation states of +IV, +II, and -II may be assigned to U, Mn, and S, respectively. U—S distances have the following ranges: U1—S: 2.7546 (19) Å to 2.9509 (3) Å; U2—S: 2.735 (3) Å to 2.8559 (4) Å; U3—S: 2.680 (2) Å to 3.031 (3) Å. These distances are similar to those found in the structures of the Cr and Fe analogues. Mn—S distances range from 2.398 (3) Å to 2.5168 (19) Å; these are shorter than a typical high-spin six-coordinate Mn^{II} —S distance of 2.66 Å, but are consistent with the typical low-spin six-coordinate Mn^{II} —S distance of 2.51 Å (Shannon, 1976). Low-spin Mn^{II} is consistent with the $^6\text{S}_{5/2}$ configuration assigned from magnetic studies (Noël & Troc, 1979).

Experimental

Black blocks of $\text{MnU}_8\text{S}_{17}$ were obtained by combining U (0.126 mmol), Mn (Johnson Matthey 99.3%, 0.126 mmol), and S (Mallinckrodt 99.6% sublimed, 0.504 mmol) in a RbCl flux (Alfa 99.8%, 1.26 mmol) with As as a mineralizer (Strem 2N, 0.126 mmol). U filings (Oak Ridge National Laboratory) were powdered by hydridization and subsequent decomposition under heat and vacuum (Bugaris & Ibers, 2008), in a modification of a previous literature method (Haneveld & Jellinek, 1969). The mixture was loaded into a carbon-coated fused-silica tube in an Ar filled glove box and then sealed under 10^{-4} Torr vacuum. The vessel was heated in a computer-controlled furnace to 1073 K in 96 h, held for 96 h, cooled to 673 K in 96 h, then cooled to 298 K in 48 h. The flux was washed off with water and surface impurities were mechanically removed from a single crystal.

Refinement

The structure was standardized by means of the program *STRUCTURE TIDY* (Gelato & Parthé, 1987). The highest peak of 3.9 (4) $\text{e}/\text{Å}^3$ is 1.33 Å from atom U2 and the deepest hole of -1.9 (4) $\text{e}/\text{Å}^3$ is 0.88 Å from atom U3.

Figures

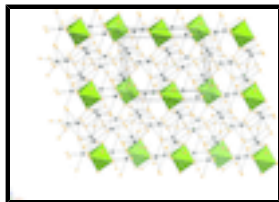


Fig. 1. A packing diagram for $\text{MnU}_8\text{S}_{17}$, viewed nearly along the b axis. U atoms are black, S atoms are yellow, and the MnSe_6 octahedra are green.

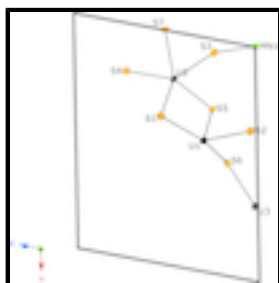


Fig. 2. The asymmetric unit of $\text{MnU}_8\text{S}_{17}$. Displacement ellipsoids at the 95% probability level are shown.

Manganese(II) octauranium(IV) heptadecasulfide

Crystal data

$\text{MnU}_8\text{S}_{17}$

$M_r = 2504.20$

Monoclinic, $C2/m$

Hall symbol: $-C\ 2y$

$a = 13.3549\ (6)\ \text{\AA}$

$b = 8.3893\ (4)\ \text{\AA}$

$c = 10.4927\ (5)\ \text{\AA}$

$\beta = 101.658\ (2)^\circ$

$V = 1151.33\ (9)\ \text{\AA}^3$

$Z = 2$

$F(000) = 2066$

$D_x = 7.223\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 8717 reflections

$\theta = 2.9\text{--}30.5^\circ$

$\mu = 58.10\ \text{mm}^{-1}$

$T = 100\ \text{K}$

Rectangular block, black

$0.15 \times 0.09 \times 0.08\ \text{mm}$

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube
graphite

φ and ω scans

Absorption correction: numerical
face indexed (*SADABS*; Sheldrick, 2008a)

$T_{\min} = 0.043$, $T_{\max} = 0.085$

8422 measured reflections

1609 independent reflections

1578 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\text{max}} = 29.1^\circ$, $\theta_{\text{min}} = 2.0^\circ$

$h = -18 \rightarrow 17$

$k = -10 \rightarrow 11$

$l = -14 \rightarrow 14$

Refinement

Refinement on F^2

Primary atom site location: structure-invariant direct
methods

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.026$$

$$wR(F^2) = 0.068$$

$$S = 2.10$$

1609 reflections

73 parameters

0 restraints

Secondary atom site location: difference Fourier map

$$w = 1/[\sigma^2(F_o^2) + (0.0161F_o^2)^2]$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 3.90 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -1.94 \text{ e } \text{\AA}^{-3}$$

Extinction correction: *SHELXL97* (Sheldrick, 2008a),

$$F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$$

Extinction coefficient: 0.00082 (4)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
U1	0.44179 (2)	0.25549 (3)	0.29633 (3)	0.00531 (11)
U2	0.20440 (3)	0.0000	0.45685 (4)	0.00510 (12)
U3	0.68285 (3)	0.0000	0.02025 (4)	0.00541 (12)
Mn1	0.0000	0.0000	0.0000	0.0058 (5)
S1	0.12730 (15)	0.3047 (2)	0.46870 (18)	0.0065 (4)
S2	0.36424 (15)	0.3062 (2)	0.03492 (18)	0.0064 (4)
S3	0.0598 (2)	0.0000	0.2314 (3)	0.0066 (5)
S4	0.2086 (2)	0.0000	0.7197 (3)	0.0061 (5)
S5	0.3030 (2)	0.0000	0.2463 (3)	0.0058 (5)
S6	0.5211 (2)	0.0000	0.1694 (3)	0.0061 (5)
S7	0.0000	0.0000	0.5000	0.0056 (7)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
U1	0.0056 (2)	0.00542 (18)	0.00528 (17)	-0.00005 (10)	0.00195 (13)	0.00004 (10)
U2	0.0053 (2)	0.0054 (2)	0.0051 (2)	0.000	0.00217 (15)	0.000
U3	0.0063 (2)	0.0052 (2)	0.00499 (19)	0.000	0.00170 (16)	0.000
Mn1	0.0047 (12)	0.0068 (11)	0.0064 (10)	0.000	0.0024 (9)	0.000
S1	0.0062 (10)	0.0067 (9)	0.0069 (8)	-0.0004 (8)	0.0017 (7)	0.0000 (7)
S2	0.0064 (10)	0.0058 (9)	0.0075 (8)	-0.0008 (7)	0.0023 (7)	0.0000 (7)
S3	0.0079 (14)	0.0067 (13)	0.0057 (11)	0.000	0.0024 (10)	0.000
S4	0.0059 (14)	0.0056 (12)	0.0071 (12)	0.000	0.0018 (10)	0.000
S5	0.0059 (14)	0.0062 (12)	0.0058 (11)	0.000	0.0027 (10)	0.000
S6	0.0067 (14)	0.0059 (12)	0.0055 (12)	0.000	0.0006 (10)	0.000
S7	0.0034 (19)	0.0050 (16)	0.0091 (17)	0.000	0.0025 (14)	0.000

Geometric parameters (\AA , $^\circ$)

U1—S3 ⁱ	2.7546 (19)	U3—U1 ^{iv}	4.0205 (4)
U1—S2	2.7621 (19)	U3—U1 ^{vii}	4.0205 (4)
U1—S1 ⁱⁱ	2.802 (2)	Mn1—S3	2.398 (3)
U1—S5	2.8129 (19)	Mn1—S3 ^x	2.398 (3)
U1—S6	2.8390 (18)	Mn1—S2 ^{xi}	2.5168 (19)

supplementary materials

U1—S1 ⁱⁱⁱ	2.8469 (19)	Mn1—S2 ^{xii}	2.5168 (19)
U1—S4 ⁱⁱⁱ	2.852 (2)	Mn1—S2 ^{xiii}	2.5168 (19)
U1—S7 ⁱ	2.9509 (3)	Mn1—S2 ^{xiv}	2.5168 (19)
U1—U3 ^{iv}	4.0206 (4)	S1—U2 ⁱⁱⁱ	2.763 (2)
U1—U2 ⁱⁱⁱ	4.0951 (5)	S1—U1 ^{xiv}	2.802 (2)
U2—S3	2.735 (3)	S1—U1 ⁱⁱⁱ	2.8469 (19)
U2—S4	2.747 (3)	S2—Mn1 ⁱ	2.5168 (19)
U2—S1 ⁱⁱⁱ	2.763 (2)	S2—U3 ^{iv}	2.680 (2)
U2—S1 ^v	2.763 (2)	S2—U3 ^{xv}	2.896 (2)
U2—S1	2.768 (2)	S3—U1 ^{xi}	2.7546 (19)
U2—S1 ^{vi}	2.768 (2)	S3—U1 ^{xiv}	2.7546 (19)
U2—S5	2.791 (3)	S4—U3 ^{viii}	2.819 (3)
U2—S7	2.8559 (4)	S4—U1 ^v	2.852 (2)
U2—U1 ⁱⁱⁱ	4.0952 (5)	S4—U1 ⁱⁱⁱ	2.852 (2)
U2—U1 ^v	4.0952 (5)	S5—U1 ^{vi}	2.8130 (19)
U3—S2 ^{iv}	2.680 (2)	S5—U3 ^{iv}	2.841 (3)
U3—S2 ^{vii}	2.680 (2)	S6—U1 ^{vi}	2.8390 (18)
U3—S4 ^{viii}	2.819 (3)	S6—U3 ^{iv}	3.031 (3)
U3—S5 ^{iv}	2.841 (3)	S7—U2 ^{xvi}	2.8559 (4)
U3—S2 ⁱⁱ	2.896 (2)	S7—U1 ^{xi}	2.9509 (3)
U3—S2 ^{ix}	2.896 (2)	S7—U1 ⁱⁱⁱ	2.9509 (3)
U3—S6	2.912 (3)	S7—U1 ^v	2.9509 (3)
U3—S6 ^{iv}	3.031 (3)	S7—U1 ^{xiv}	2.9509 (3)
S3 ⁱ —U1—S2	76.01 (7)	S2 ^{vii} —U3—S2 ^{ix}	134.99 (5)
S3 ⁱ —U1—S1 ⁱⁱ	79.55 (7)	S4 ^{viii} —U3—S2 ^{ix}	71.79 (6)
S2—U1—S1 ⁱⁱ	140.70 (6)	S5 ^{iv} —U3—S2 ^{ix}	80.27 (6)
S3 ⁱ —U1—S5	155.43 (8)	S2 ⁱⁱ —U3—S2 ^{ix}	68.33 (8)
S2—U1—S5	80.28 (7)	S2 ^{iv} —U3—S6	87.07 (5)
S1 ⁱⁱ —U1—S5	116.52 (6)	S2 ^{vii} —U3—S6	87.07 (5)
S3 ⁱ —U1—S6	99.20 (7)	S4 ^{viii} —U3—S6	76.83 (8)
S2—U1—S6	75.58 (7)	S5 ^{iv} —U3—S6	137.15 (8)
S1 ⁱⁱ —U1—S6	78.56 (7)	S2 ⁱⁱ —U3—S6	132.44 (5)
S5—U1—S6	68.37 (7)	S2 ^{ix} —U3—S6	132.44 (5)
S3 ⁱ —U1—S1 ⁱⁱⁱ	130.39 (7)	S2 ^{iv} —U3—S6 ^{iv}	73.60 (4)
S2—U1—S1 ⁱⁱⁱ	139.93 (6)	S2 ^{vii} —U3—S6 ^{iv}	73.60 (4)
S1 ⁱⁱ —U1—S1 ⁱⁱⁱ	78.93 (6)	S4 ^{viii} —U3—S6 ^{iv}	148.59 (8)
S5—U1—S1 ⁱⁱⁱ	73.14 (7)	S5 ^{iv} —U3—S6 ^{iv}	65.38 (8)
S6—U1—S1 ⁱⁱⁱ	119.29 (6)	S2 ⁱⁱ —U3—S6 ^{iv}	131.75 (5)
S3 ⁱ —U1—S4 ⁱⁱⁱ	83.16 (6)	S2 ^{ix} —U3—S6 ^{iv}	131.75 (5)
S2—U1—S4 ⁱⁱⁱ	73.29 (7)	S6—U3—S6 ^{iv}	71.77 (8)

S1 ⁱⁱ —U1—S4 ⁱⁱⁱ	133.46 (7)	S2 ^{iv} —U3—U1 ^{iv}	43.17 (4)
S5—U1—S4 ⁱⁱⁱ	96.17 (6)	S2 ^{vii} —U3—U1 ^{iv}	107.02 (4)
S6—U1—S4 ⁱⁱⁱ	147.19 (8)	S4 ^{viii} —U3—U1 ^{iv}	147.748 (6)
S1 ⁱⁱⁱ —U1—S4 ⁱⁱⁱ	80.27 (7)	S5 ^{iv} —U3—U1 ^{iv}	44.40 (4)
S3 ⁱ —U1—S7 ⁱ	65.25 (5)	S2 ⁱⁱ —U3—U1 ^{iv}	123.06 (4)
S2—U1—S7 ⁱ	127.10 (4)	S2 ^{ix} —U3—U1 ^{iv}	86.95 (4)
S1 ⁱⁱ —U1—S7 ⁱ	65.71 (4)	S6—U3—U1 ^{iv}	102.46 (5)
S5—U1—S7 ⁱ	137.07 (6)	S6 ^{iv} —U3—U1 ^{iv}	44.81 (3)
S6—U1—S7 ⁱ	142.74 (6)	S2 ^{iv} —U3—U1 ^{vii}	107.02 (4)
S1 ⁱⁱⁱ —U1—S7 ⁱ	65.18 (4)	S2 ^{vii} —U3—U1 ^{vii}	43.17 (4)
S4 ⁱⁱⁱ —U1—S7 ⁱ	67.79 (5)	S4 ^{viii} —U3—U1 ^{vii}	147.748 (6)
S3 ⁱ —U1—U3 ^{iv}	110.77 (6)	S5 ^{iv} —U3—U1 ^{vii}	44.40 (4)
S2—U1—U3 ^{iv}	41.58 (4)	S2 ⁱⁱ —U3—U1 ^{vii}	86.95 (4)
S1 ⁱⁱ —U1—U3 ^{iv}	126.98 (4)	S2 ^{ix} —U3—U1 ^{vii}	123.06 (4)
S5—U1—U3 ^{iv}	44.95 (6)	S6—U3—U1 ^{vii}	102.46 (5)
S6—U1—U3 ^{iv}	48.80 (6)	S6 ^{iv} —U3—U1 ^{vii}	44.81 (3)
S1 ⁱⁱⁱ —U1—U3 ^{iv}	117.93 (4)	U1 ^{iv} —U3—U1 ^{vii}	64.432 (10)
S4 ⁱⁱⁱ —U1—U3 ^{iv}	99.54 (5)	S3—Mn1—S3 ^x	180.0
S7 ⁱ —U1—U3 ^{iv}	166.765 (10)	S3—Mn1—S2 ^{xi}	87.41 (7)
S3 ⁱ —U1—U2 ⁱⁱⁱ	98.80 (5)	S3 ^x —Mn1—S2 ^{xi}	92.59 (7)
S2—U1—U2 ⁱⁱⁱ	114.83 (4)	S3—Mn1—S2 ^{xii}	92.59 (7)
S1 ⁱⁱ —U1—U2 ⁱⁱⁱ	98.88 (4)	S3 ^x —Mn1—S2 ^{xii}	87.41 (7)
S5—U1—U2 ⁱⁱⁱ	96.86 (5)	S2 ^{xi} —Mn1—S2 ^{xii}	180.00 (11)
S6—U1—U2 ⁱⁱⁱ	161.01 (4)	S3—Mn1—S2 ^{xiii}	92.59 (7)
S1 ⁱⁱⁱ —U1—U2 ⁱⁱⁱ	42.42 (4)	S3 ^x —Mn1—S2 ^{xiii}	87.41 (7)
S4 ⁱⁱⁱ —U1—U2 ⁱⁱⁱ	42.01 (5)	S2 ^{xi} —Mn1—S2 ^{xiii}	99.50 (9)
S7 ⁱ —U1—U2 ⁱⁱⁱ	44.218 (7)	S2 ^{xii} —Mn1—S2 ^{xiii}	80.50 (9)
U3 ^{iv} —U1—U2 ⁱⁱⁱ	128.230 (12)	S3—Mn1—S2 ^{xiv}	87.41 (7)
S3—U2—S4	137.38 (8)	S3 ^x —Mn1—S2 ^{xiv}	92.59 (7)
S3—U2—S1 ⁱⁱⁱ	129.53 (6)	S2 ^{xi} —Mn1—S2 ^{xiv}	80.50 (9)
S4—U2—S1 ⁱⁱⁱ	82.29 (6)	S2 ^{xii} —Mn1—S2 ^{xiv}	99.50 (9)
S3—U2—S1 ^v	129.53 (6)	S2 ^{xiii} —Mn1—S2 ^{xiv}	180.00 (9)
S4—U2—S1 ^v	82.29 (6)	U2 ⁱⁱⁱ —S1—U2	105.75 (7)
S1 ⁱⁱⁱ —U2—S1 ^v	72.72 (8)	U2 ⁱⁱⁱ —S1—U1 ^{xiv}	148.40 (8)
S3—U2—S1	80.49 (5)	U2—S1—U1 ^{xiv}	95.33 (6)
S4—U2—S1	83.52 (5)	U2 ⁱⁱⁱ —S1—U1 ⁱⁱⁱ	104.35 (6)
S1 ⁱⁱⁱ —U2—S1	74.25 (7)	U2—S1—U1 ⁱⁱⁱ	93.65 (6)
S1 ^v —U2—S1	145.44 (5)	U1 ^{xiv} —S1—U1 ⁱⁱⁱ	97.35 (6)
S3—U2—S1 ^{vi}	80.49 (5)	Mn1 ⁱ —S2—U3 ^{iv}	137.09 (8)
S4—U2—S1 ^{vi}	83.52 (5)	Mn1 ⁱ —S2—U1	96.17 (6)
S1 ⁱⁱⁱ —U2—S1 ^{vi}	145.44 (5)	U3 ^{iv} —S2—U1	95.25 (6)

supplementary materials

S1 ^v —U2—S1 ^{vi}	74.25 (7)	Mn1 ⁱ —S2—U3 ^{xv}	104.41 (7)
S1—U2—S1 ^{vi}	134.89 (9)	U3 ^{iv} —S2—U3 ^{xv}	111.68 (7)
S3—U2—S5	71.28 (8)	U1—S2—U3 ^{xv}	106.32 (6)
S4—U2—S5	151.35 (9)	Mn1—S3—U2	155.27 (12)
S1 ⁱⁱⁱ —U2—S5	74.75 (6)	Mn1—S3—U1 ^{xi}	99.23 (8)
S1 ^v —U2—S5	74.75 (6)	U2—S3—U1 ^{xi}	97.20 (7)
S1—U2—S5	105.95 (4)	Mn1—S3—U1 ^{xiv}	99.23 (8)
S1 ^{vi} —U2—S5	105.95 (4)	U2—S3—U1 ^{xiv}	97.20 (7)
S3—U2—S7	66.83 (6)	U1 ^{xi} —S3—U1 ^{xiv}	96.26 (9)
S4—U2—S7	70.55 (6)	U2—S4—U3 ^{viii}	150.90 (11)
S1 ⁱⁱⁱ —U2—S7	134.75 (4)	U2—S4—U1 ^v	93.99 (7)
S1 ^v —U2—S7	134.75 (4)	U3 ^{viii} —S4—U1 ^v	105.98 (7)
S1—U2—S7	67.46 (4)	U2—S4—U1 ⁱⁱⁱ	93.99 (7)
S1 ^{vi} —U2—S7	67.46 (4)	U3 ^{viii} —S4—U1 ⁱⁱⁱ	105.98 (7)
S5—U2—S7	138.10 (6)	U1 ^v —S4—U1 ⁱⁱⁱ	91.99 (8)
S3—U2—U1 ⁱⁱⁱ	101.73 (5)	U2—S5—U1	104.51 (7)
S4—U2—U1 ⁱⁱⁱ	44.00 (4)	U2—S5—U1 ^{vi}	104.51 (7)
S1 ⁱⁱⁱ —U2—U1 ⁱⁱⁱ	89.36 (4)	U1—S5—U1 ^{vi}	99.28 (9)
S1 ^v —U2—U1 ⁱⁱⁱ	125.64 (4)	U2—S5—U3 ^{iv}	156.22 (12)
S1—U2—U1 ⁱⁱⁱ	43.93 (4)	U1—S5—U3 ^{iv}	90.65 (7)
S1 ^{vi} —U2—U1 ⁱⁱⁱ	101.83 (4)	U1 ^{vi} —S5—U3 ^{iv}	90.65 (7)
S5—U2—U1 ⁱⁱⁱ	149.546 (11)	U1 ^{vi} —S6—U1	98.05 (8)
S7—U2—U1 ⁱⁱⁱ	46.101 (7)	U1 ^{vi} —S6—U3	129.76 (5)
S3—U2—U1 ^v	101.73 (5)	U1—S6—U3	129.76 (5)
S4—U2—U1 ^v	44.00 (4)	U1 ^{vi} —S6—U3 ^{iv}	86.39 (7)
S1 ⁱⁱⁱ —U2—U1 ^v	125.64 (4)	U1—S6—U3 ^{iv}	86.39 (7)
S1 ^v —U2—U1 ^v	89.36 (4)	U3—S6—U3 ^{iv}	108.23 (8)
S1—U2—U1 ^v	101.83 (4)	U2—S7—U2 ^{xvi}	180.0
S1 ^{vi} —U2—U1 ^v	43.93 (4)	U2—S7—U1 ^{xi}	90.317 (8)
S5—U2—U1 ^v	149.546 (10)	U2 ^{xvi} —S7—U1 ^{xi}	89.682 (8)
S7—U2—U1 ^v	46.101 (7)	U2—S7—U1 ⁱⁱⁱ	89.683 (8)
U1 ⁱⁱⁱ —U2—U1 ^v	60.118 (10)	U2 ^{xvi} —S7—U1 ⁱⁱⁱ	90.318 (8)
S2 ^{iv} —U3—S2 ^{vii}	146.88 (9)	U1 ^{xi} —S7—U1 ⁱⁱⁱ	180.0
S2 ^{iv} —U3—S4 ^{viii}	105.15 (4)	U2—S7—U1 ^v	89.683 (8)
S2 ^{vii} —U3—S4 ^{viii}	105.15 (4)	U2 ^{xvi} —S7—U1 ^v	90.318 (8)
S2 ^{iv} —U3—S5 ^{iv}	81.18 (4)	U1 ^{xi} —S7—U1 ^v	91.925 (11)
S2 ^{vii} —U3—S5 ^{iv}	81.18 (4)	U1 ⁱⁱⁱ —S7—U1 ^v	88.075 (11)
S4 ^{viii} —U3—S5 ^{iv}	146.02 (8)	U2—S7—U1 ^{xiv}	90.317 (8)
S2 ^{iv} —U3—S2 ⁱⁱ	134.99 (5)	U2 ^{xvi} —S7—U1 ^{xiv}	89.682 (8)
S2 ^{vii} —U3—S2 ⁱⁱ	68.31 (7)	U1 ^{xi} —S7—U1 ^{xiv}	88.075 (11)
S4 ^{viii} —U3—S2 ⁱⁱ	71.79 (6)	U1 ⁱⁱⁱ —S7—U1 ^{xiv}	91.925 (11)

S5^{iv}—U3—S2ⁱⁱ 80.27 (6) U1^v—S7—U1^{xiv} 180.0
 S2^{iv}—U3—S2^{ix} 68.31 (7)

Symmetry codes: (i) $x+1/2, y+1/2, z$; (ii) $x+1/2, -y+1/2, z$; (iii) $-x+1/2, -y+1/2, -z+1$; (iv) $-x+1, -y, -z$; (v) $-x+1/2, y-1/2, -z+1$; (vi) $x, -y, z$; (vii) $-x+1, y, -z$; (viii) $-x+1, -y, -z+1$; (ix) $x+1/2, y-1/2, z$; (x) $-x, -y, -z$; (xi) $x-1/2, y-1/2, z$; (xii) $-x+1/2, -y+1/2, -z$; (xiii) $-x+1/2, y-1/2, -z$; (xiv) $x-1/2, -y+1/2, z$; (xv) $x-1/2, y+1/2, z$; (xvi) $-x, -y, -z+1$.

Fig. 1

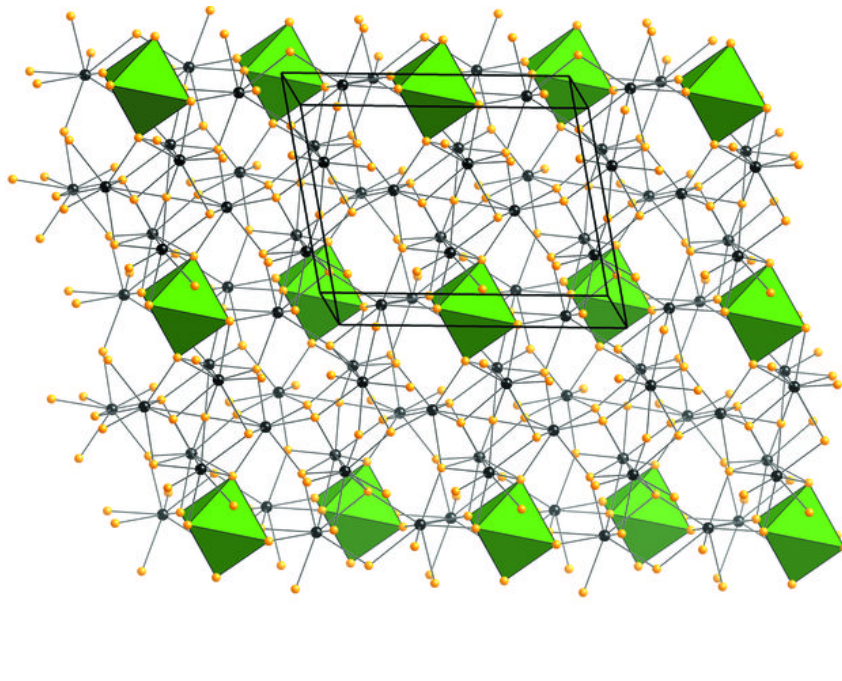


Fig. 2

