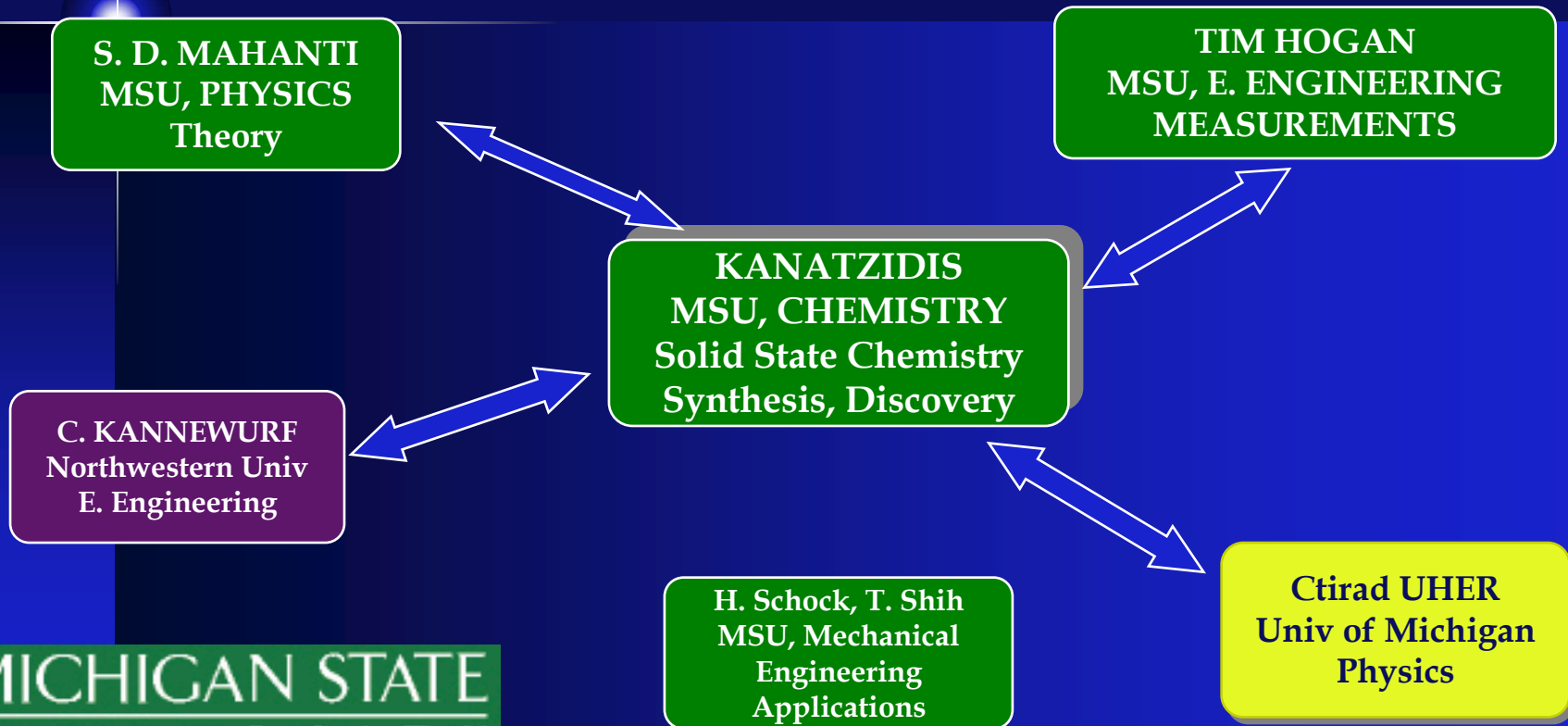


# Solid State Chemistry Meets Physcis: Thermoelectric Materials

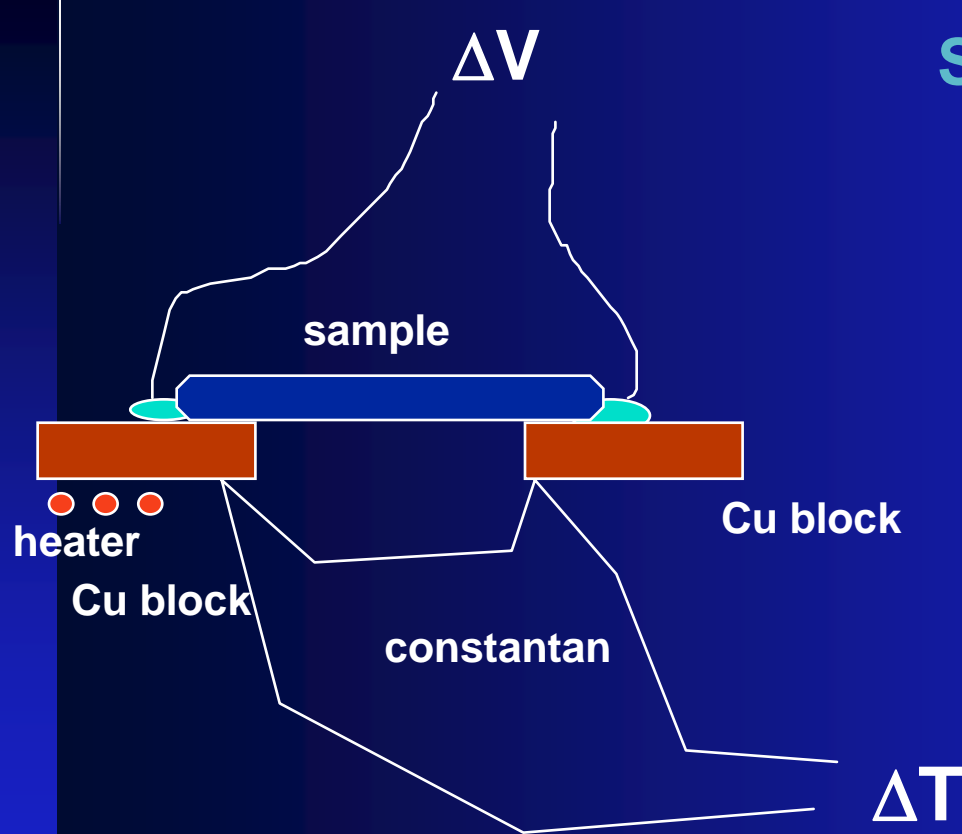
## AN INTERDISCIPLINARY COLLABORATION



# Michigan State University...



# THERMOELECTRIC POWER (Seebeck Coefficient)



$$S_{\text{measured}} = \frac{\Delta V}{\Delta T}$$

$$S_{\text{measured}} = S_{\text{sample}} - S_{\text{Cu}}$$

**zero current technique:**  
extremely useful probe  
for investigation of  
intrinsic conduction in  
granular or polycrystalline  
materials

# Thermoelectric Applications

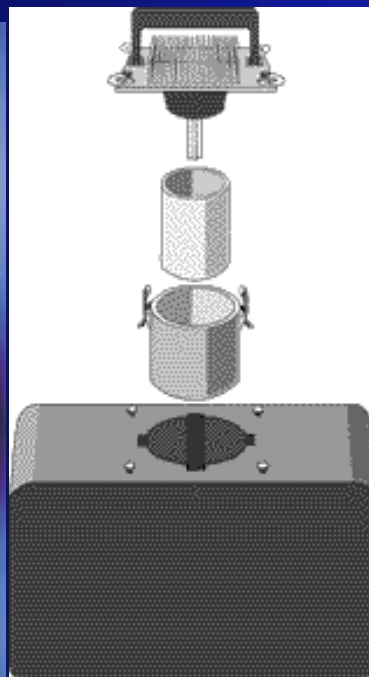
PRESENTS  
**The Cool World**

[www.tellurex.com](http://www.tellurex.com)

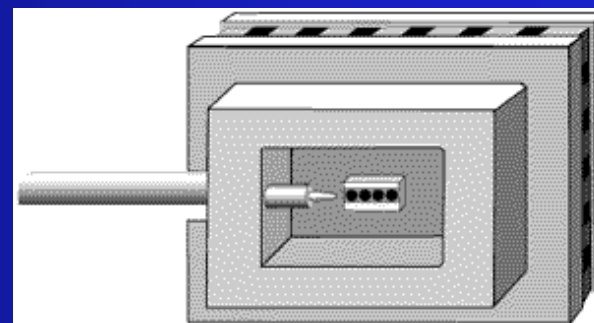
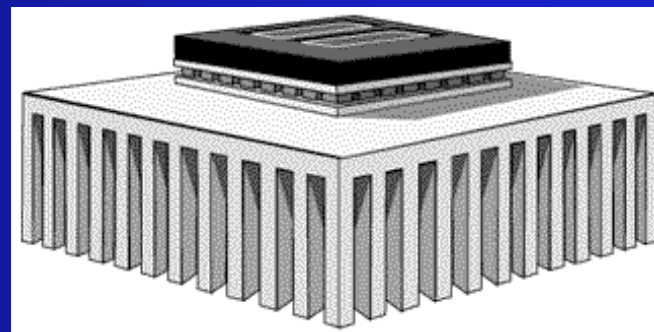
Microprocessor  
Cooler



Beverage cooler

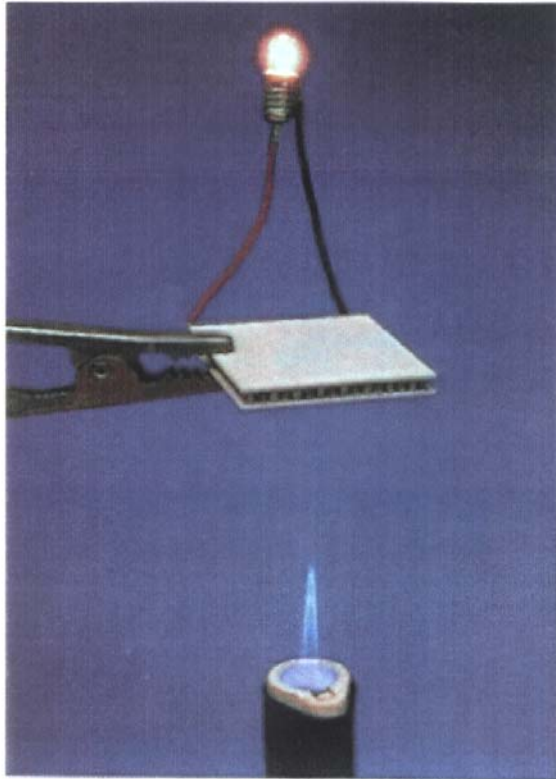


Biological samples



LASER diode  
Cooler

# Heat to Electric Energy



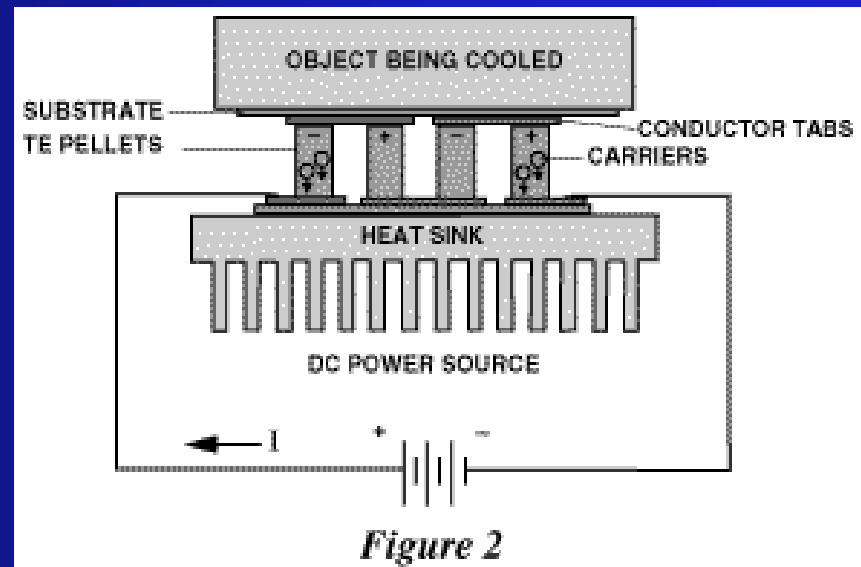
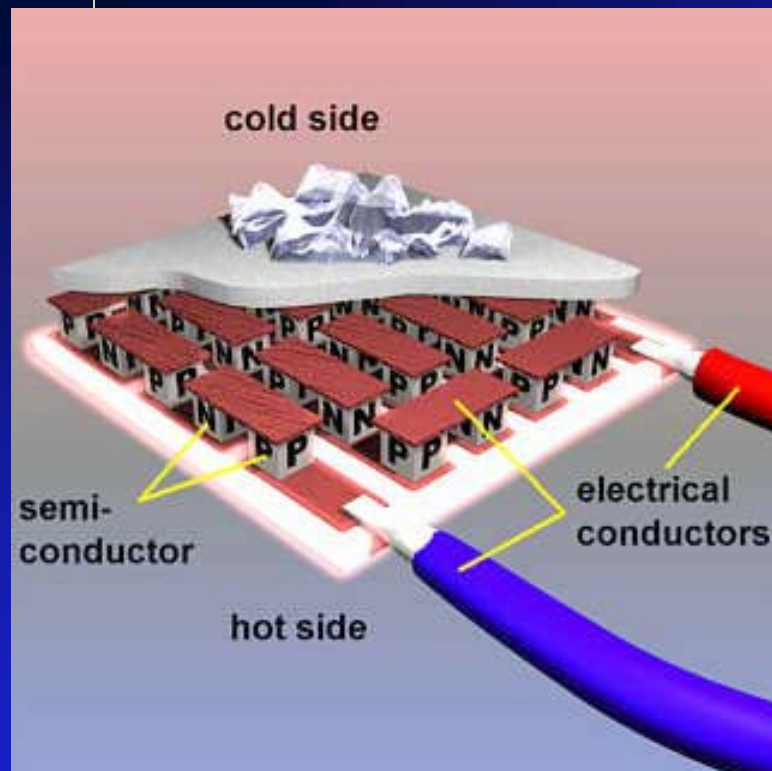
**Electrical  
Power Generation**

Up to 20% conversion efficiency with  
right materials

# Thermoelectric Benefits

- TE coolers have no moving parts, need substantially less maintenance.
- Life-testing has shown the capability of TE devices to exceed 200,000 hrs. of steady state operation.
- TE coolers contain no chlorofluorocarbons.
- Temperature control to within fractions of a degree can be maintained using TE devices.
- TE coolers function in environments that are too severe, too sensitive, or too small for conventional refrigeration.
- TE coolers are not position-dependent.

# How does it work?



# Thermoelectric applications

- Air conditioning (distributed, environmentally friendly)
- Spot cooling of electronic chips, superconductors etc.
- Thermal suits for fire-fighting, soldier etc
- Waste heat recovery (automobiles, utilities etc)
- Geothermal power generation



# Figure of Merit

electrical conductivity

thermopower

$$ZT = \frac{\sigma \cdot S^2}{K_{total}} \cdot T$$

Total thermal conductivity

Power factor

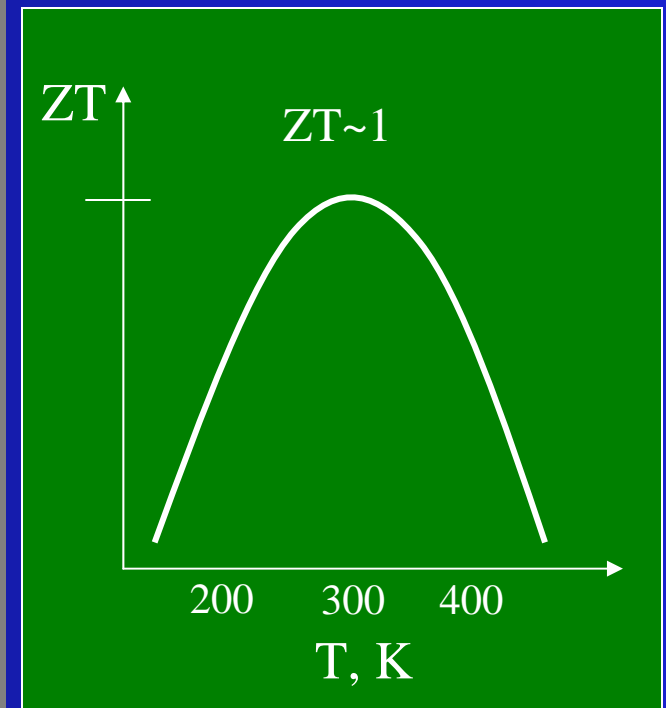
$$\sigma \cdot S^2$$

# Today's situation

- The most efficient materials today is  $\text{Bi}_2\text{Te}_3$  alloy
- $ZT \sim 0.8-1.0$
- Further improvements on  $\text{Bi}_2\text{Te}_3$  are not expected.
- New materials are needed

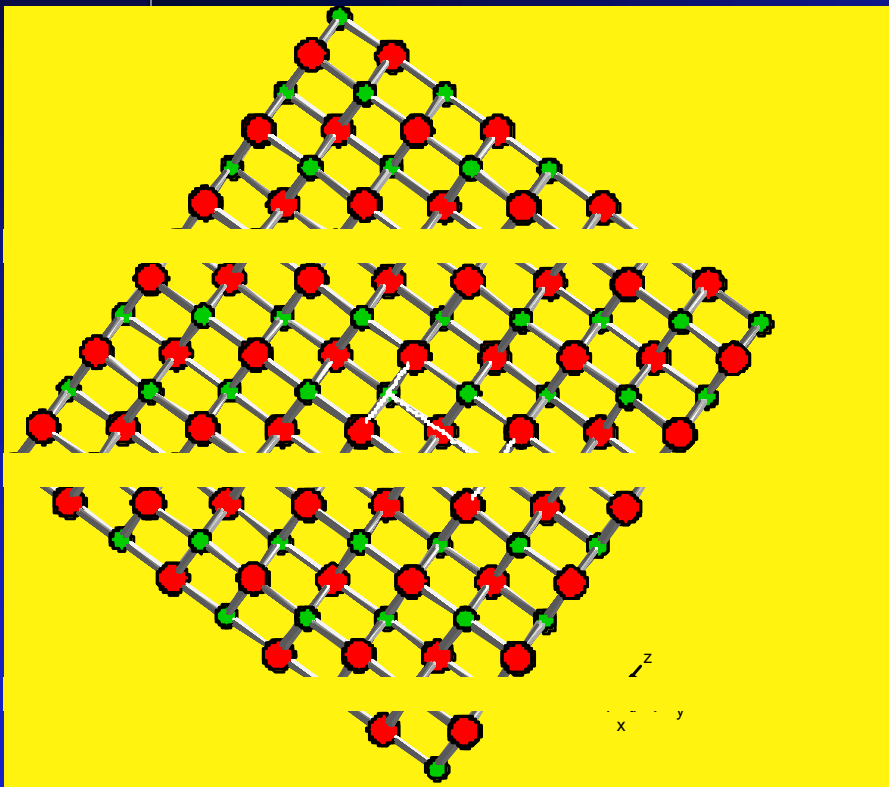
# Thermoelectric Properties of Optimized $\text{Bi}_2\text{Te}_3$ (e.g. $\text{Bi}_{2-x}\text{Sb}_x\text{Te}_3$ , $\text{Bi}_2\text{Te}_{3-x}\text{Se}_x$ ) at Room Temperature

- $S \sim \pm 220 \mu\text{V/K}$
- $\sigma \sim 950 \text{ S/cm}$
- $\rho = 1/\sigma \sim 1.1 \text{ m}\Omega\cdot\text{cm}$
- $\kappa \sim 1.5 \text{ W/m}\cdot\text{K}$
  
- $ZT \sim 1!$

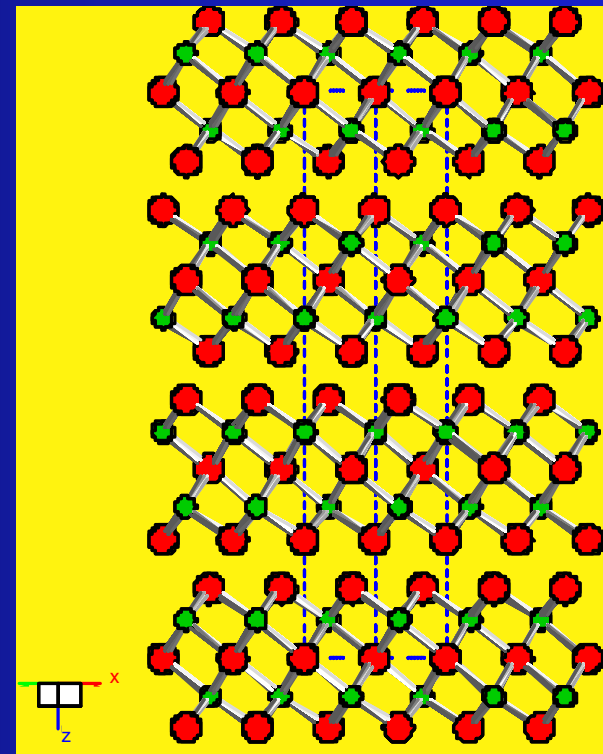


# Structure of $\text{Bi}_2\text{Te}_3$ and NaCl

NaCl

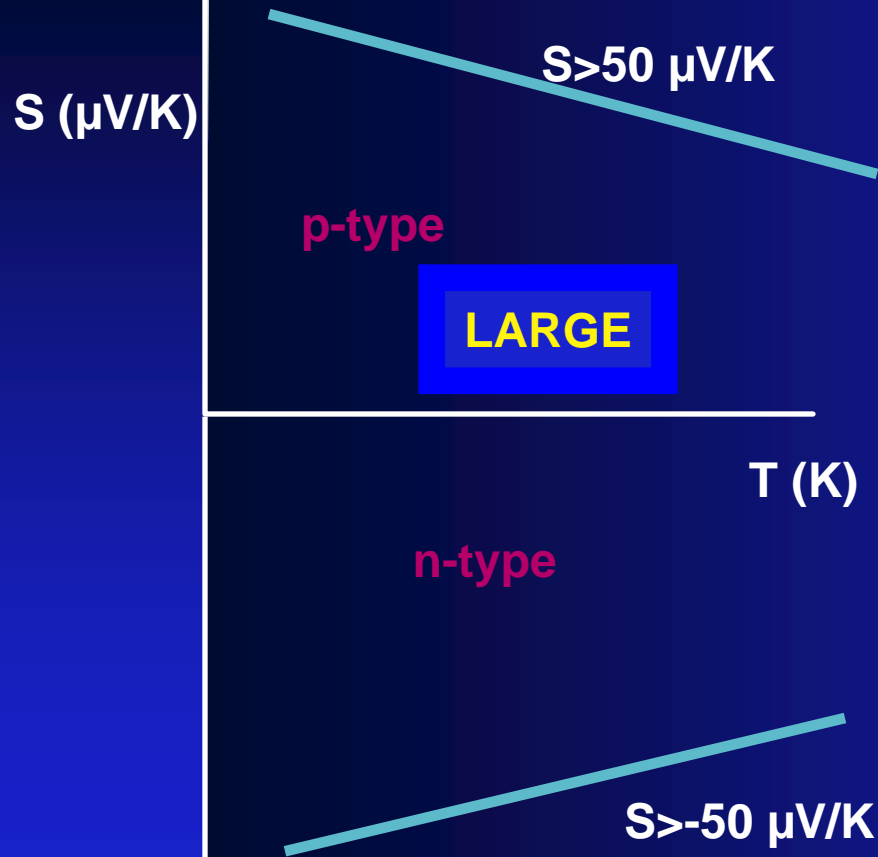


$\text{Bi}_2\text{Te}_3$  defect NaCl

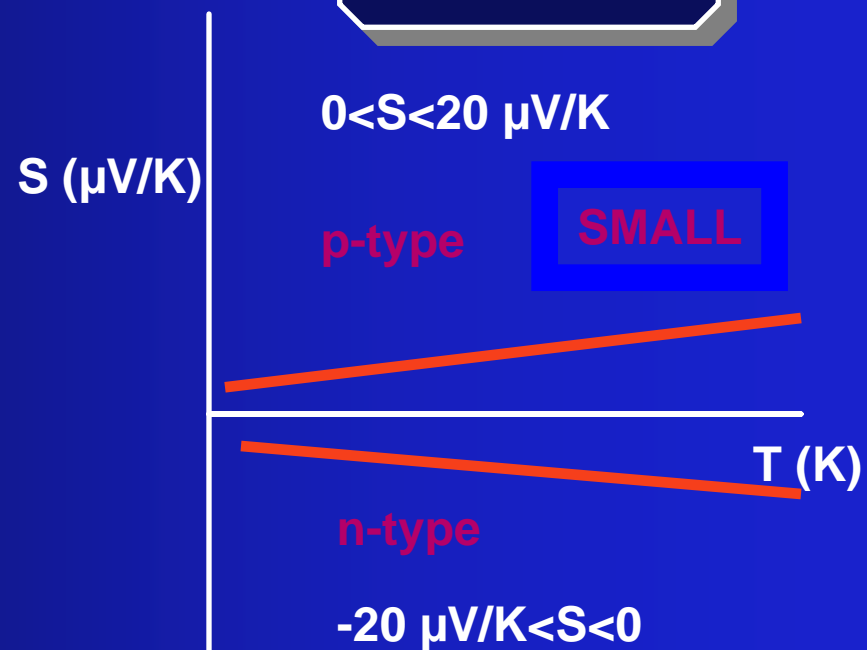


# TYPICAL BEHAVIOR OF MATERIALS

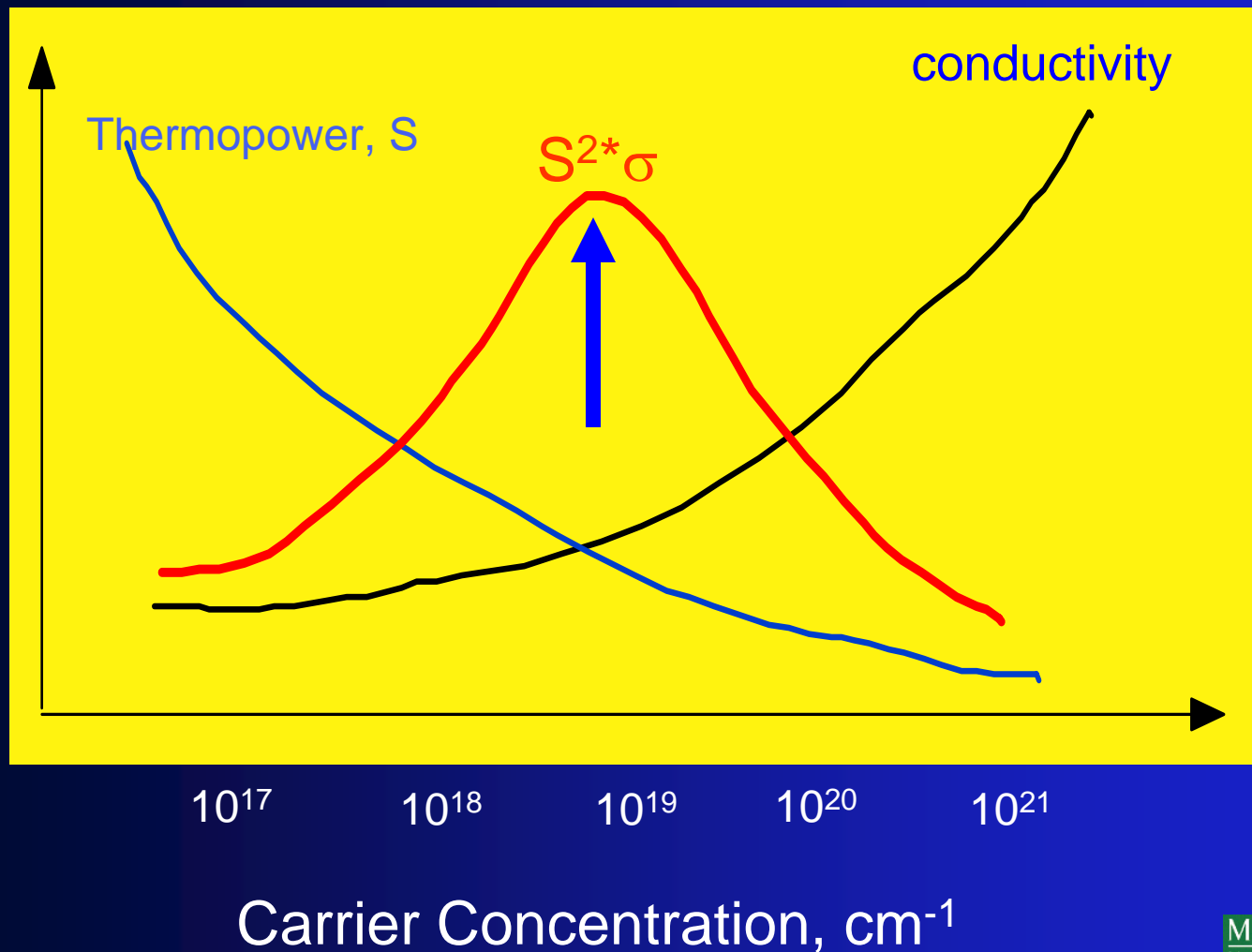
semiconductor-like



metal-like



# Power Factor ( $S^2 \cdot \sigma$ ) vs Carrier Concentration

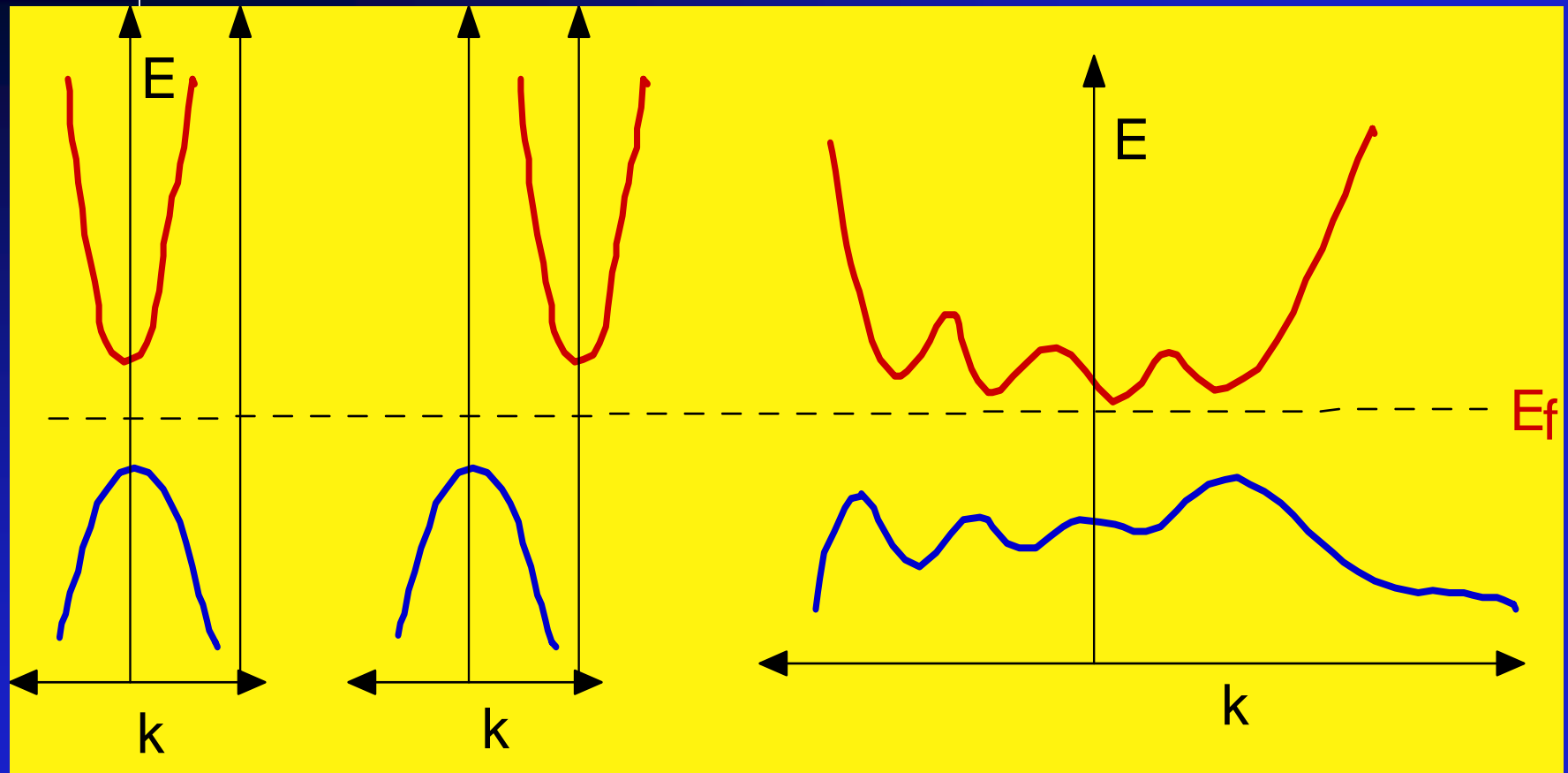


# Thermopower and Electronic Structure

$$S = \frac{\pi^2 k_B^2 T}{3e} \left. \frac{d(\ln \sigma(E))}{dE} \right|_{E = E_F} \quad \text{Mott Equation}$$

- $\sigma(E)$  is the electrical conductivity determined as a function of band filling or Fermi energy,  $E_F$ . If the electronic scattering is independent of energy,  $\sigma(E)$  is just proportional to the density of states (DOS) at  $E_F$ .
- For maximum  $S$ , a large asymmetry in the DOS and/or scattering within a few  $kT$  above and below the Fermi energy is required.

# Band structure Types



simple

simple

complex



# ZT and Band Structure

B- parameter

$$B = \frac{CT^{5/2}\gamma\sqrt{m_x m_y m_z} \mu_x}{\kappa_{latt}}$$

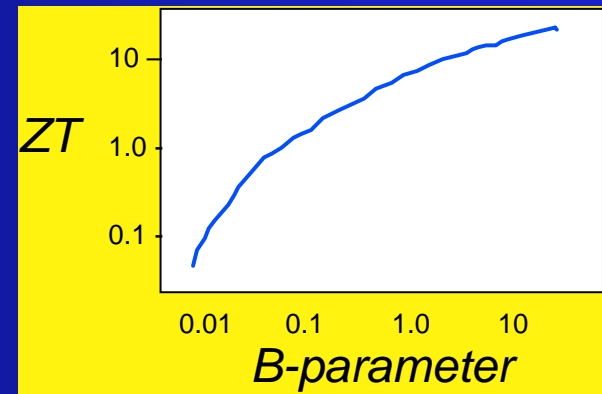
m= effective mass

$\mu$ = mobility

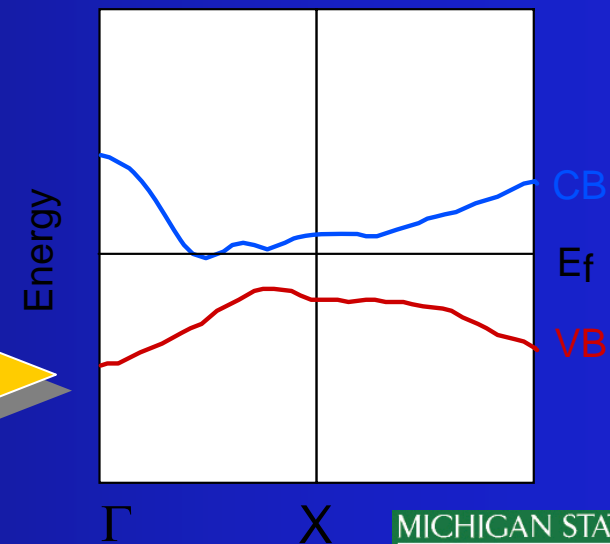
$\kappa_{latt}$ = lattice thermal conductivity

T = temperature

$\gamma$ = band degeneracy

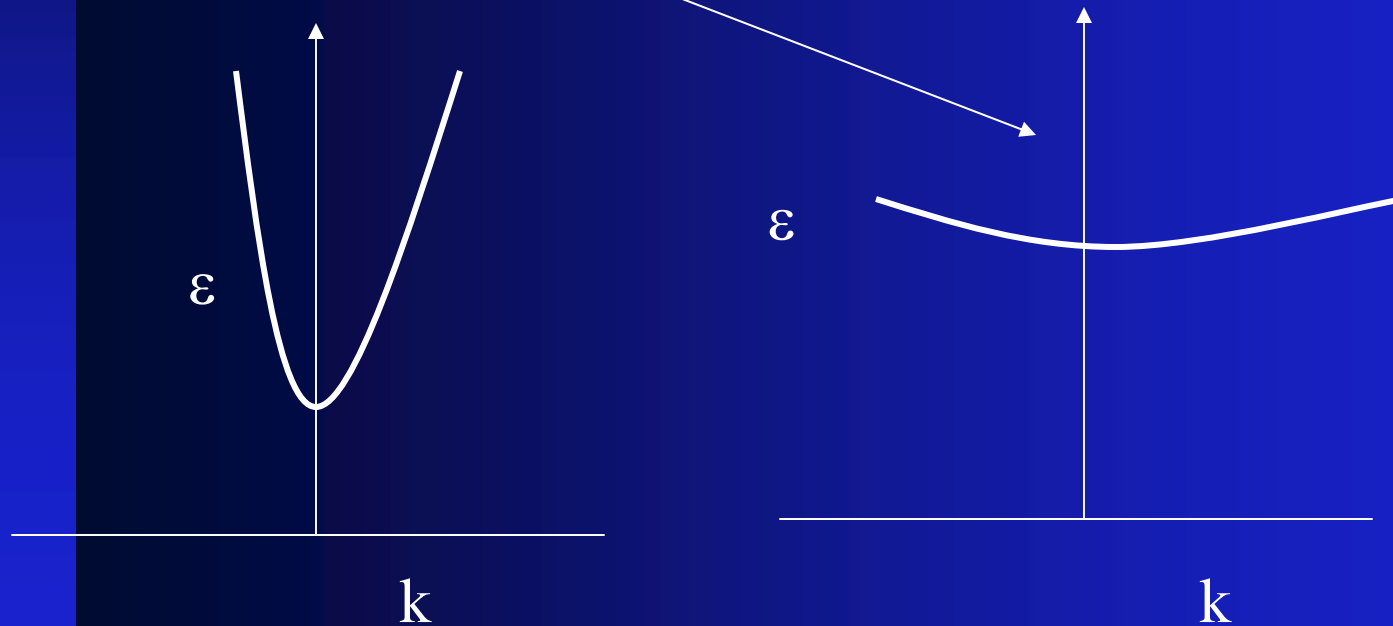


High  $\gamma$  comes with  
(a) high symmetry e.g. rhombohedral, cubic  
(b) off-center band extrema



# Desirable characteristics

- Multiple peaks and valleys in valence/conduction band
- Heavy carrier masses
  - Flat bands

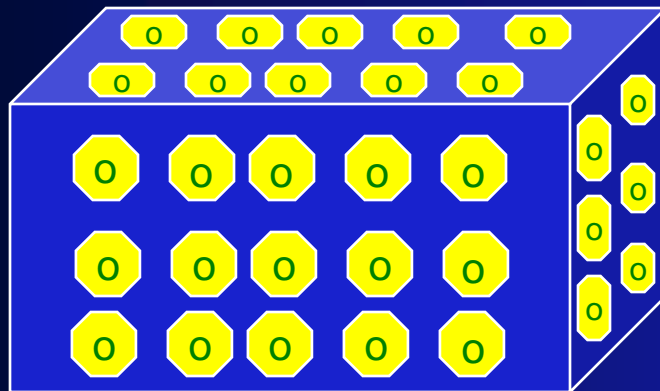


# Selection criteria for candidate materials

- Narrow band-gap semiconductors
  - For operation at room temperature
- Heavy elements
  - High mobility, low thermal conductivity
- Large unit cell, complex structure
  - low thermal conductivity
- Highly anisotropic or highly symmetric
- Complex compositions
  - low thermal conductivity, electronic structure

# Important Issue: Thermal Conductivity

- Slack's proposal: Phonon-Glass/ Electron-Crystal (PGEC)
  - Rattling Ions in the lattice: watch thermal displacement parameters



Crystalline solid

Rattling ions in cavities  
or tunnels  
scatter heat-carrying phonons

# Reaction Chemistry

Investigating the System:

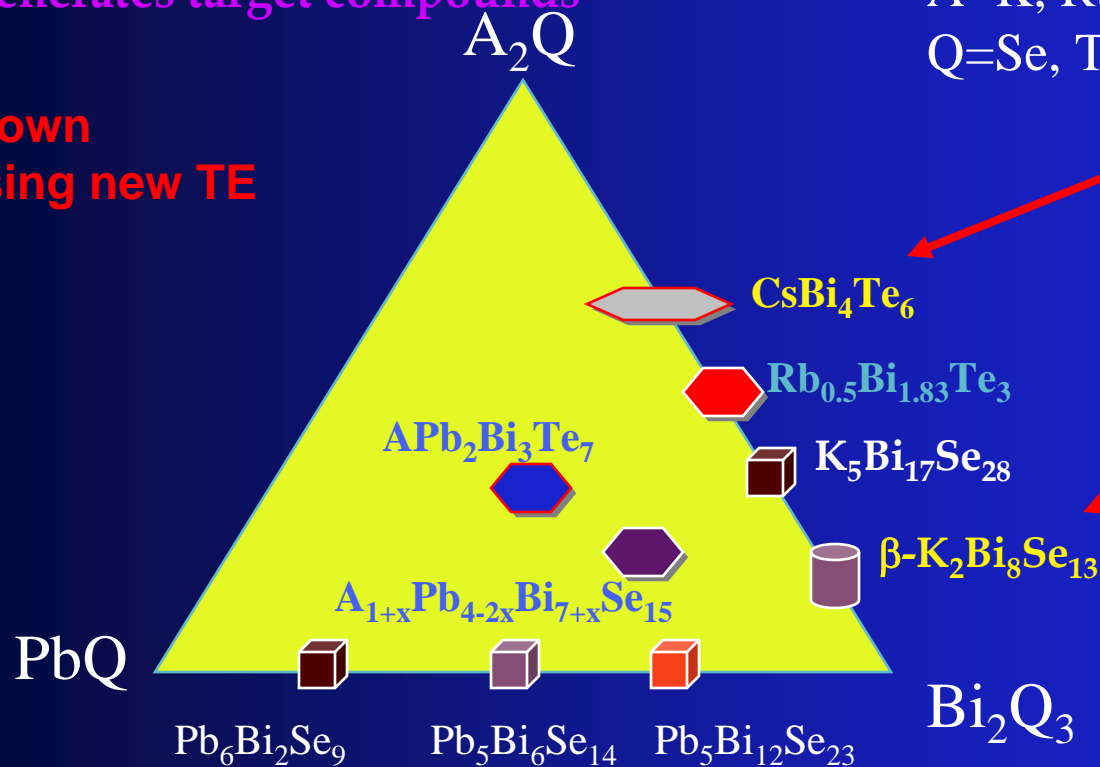


Map generates target compounds

A=K, Rb, Cs

Q=Se, Te

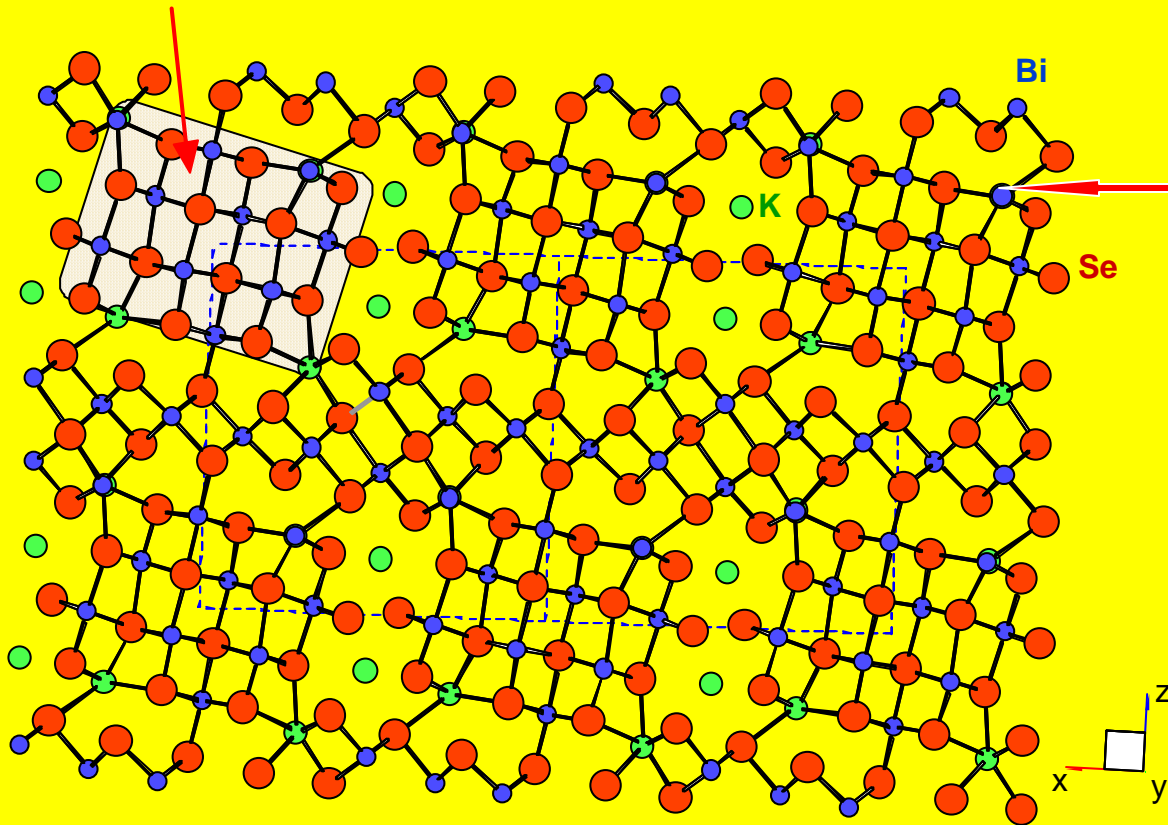
Phases shown  
are promising new TE  
Materials



# $K_2Bi_8Se_{13}$

## $\beta$ - $K_2Bi_8Se_{13}$

this block is a chunk  
from the NaCl lattice

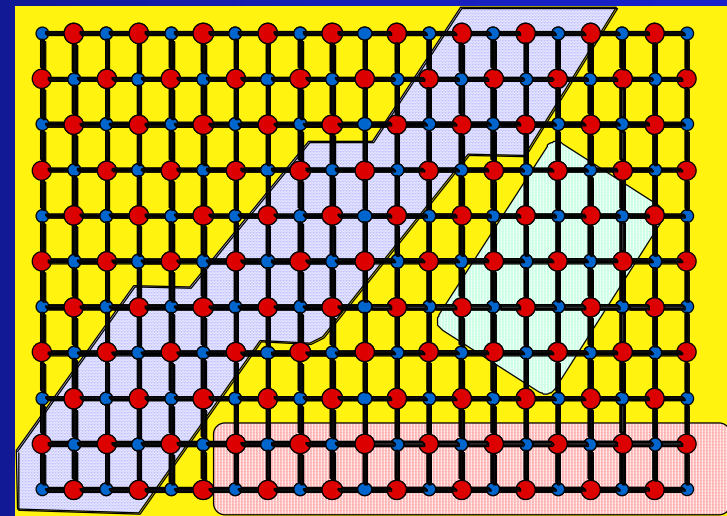
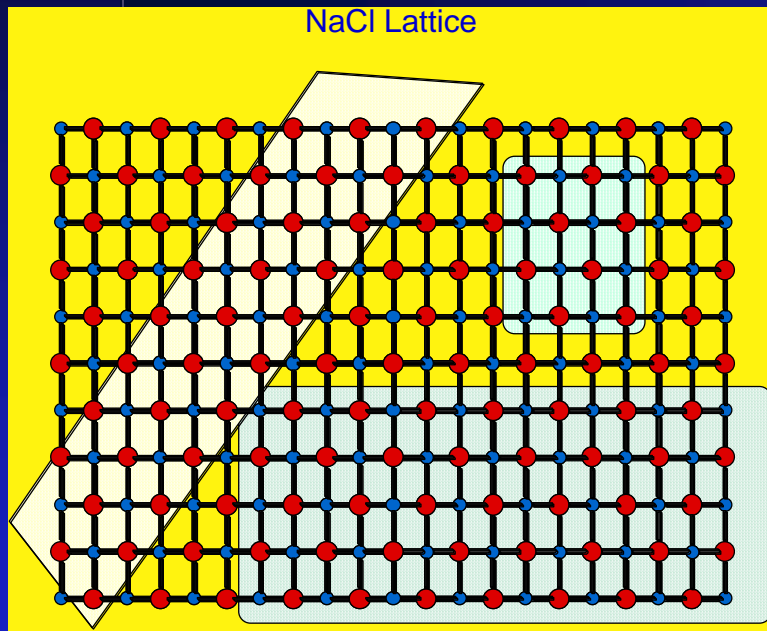


K ions possess ~2x the  
thermal displacement  
parameter  
of the Bi/Se framework

8,9-coordinate sites  
K, Bi

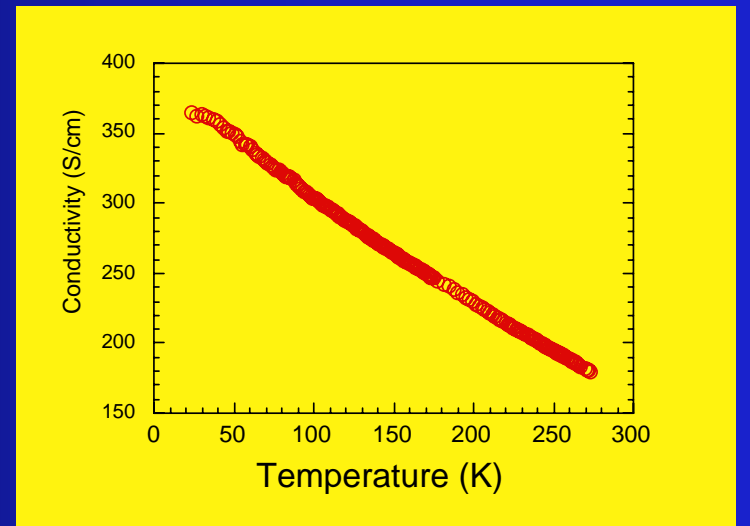
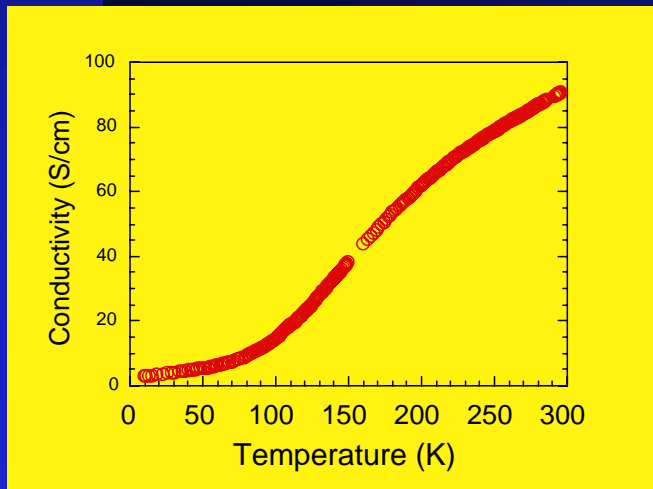
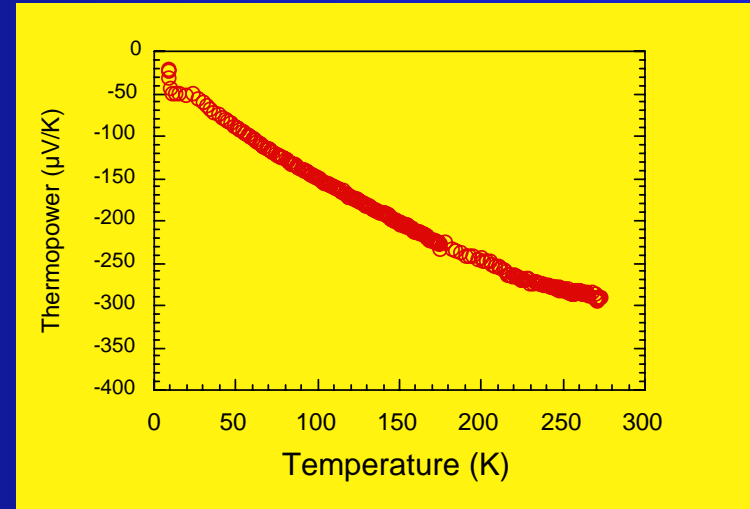
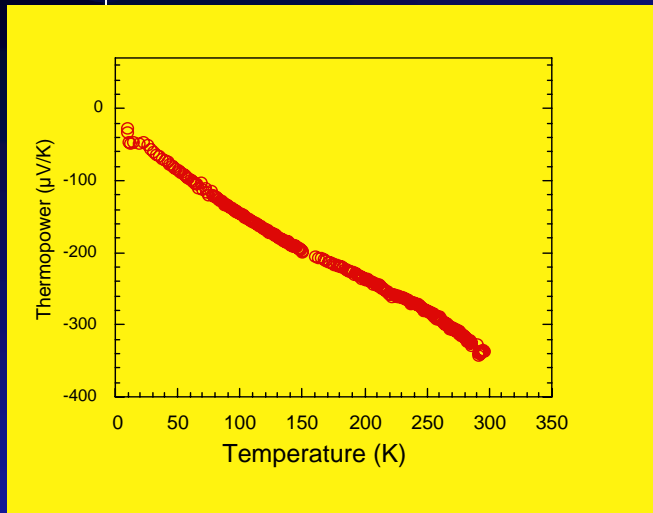
# NaCl Structure: The Basic “Raw” Material

“Modules” are cut out of NaCl stock



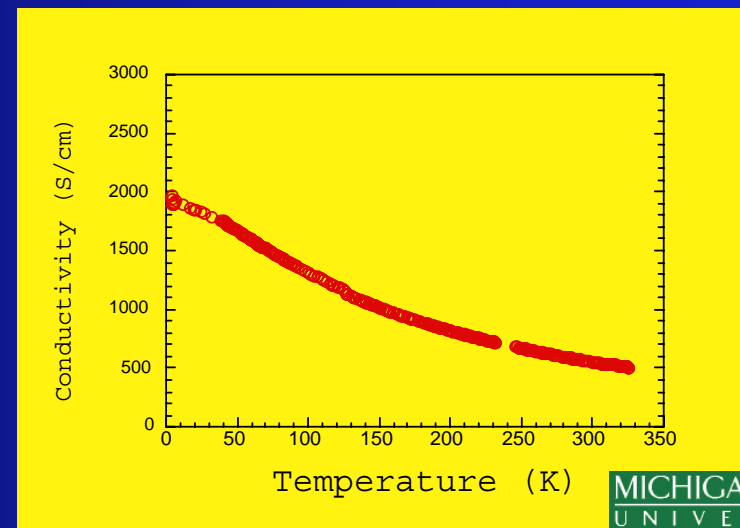
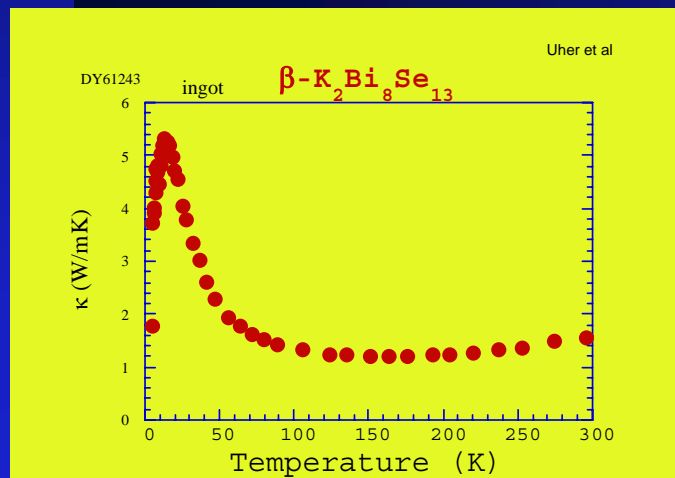
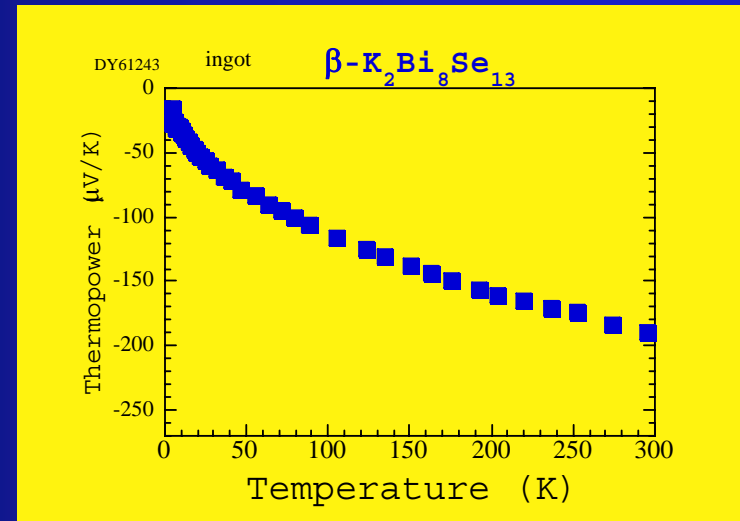
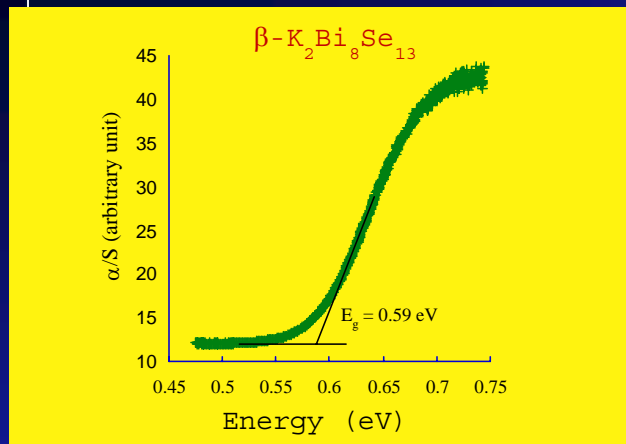
# $\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$

Sn Doped

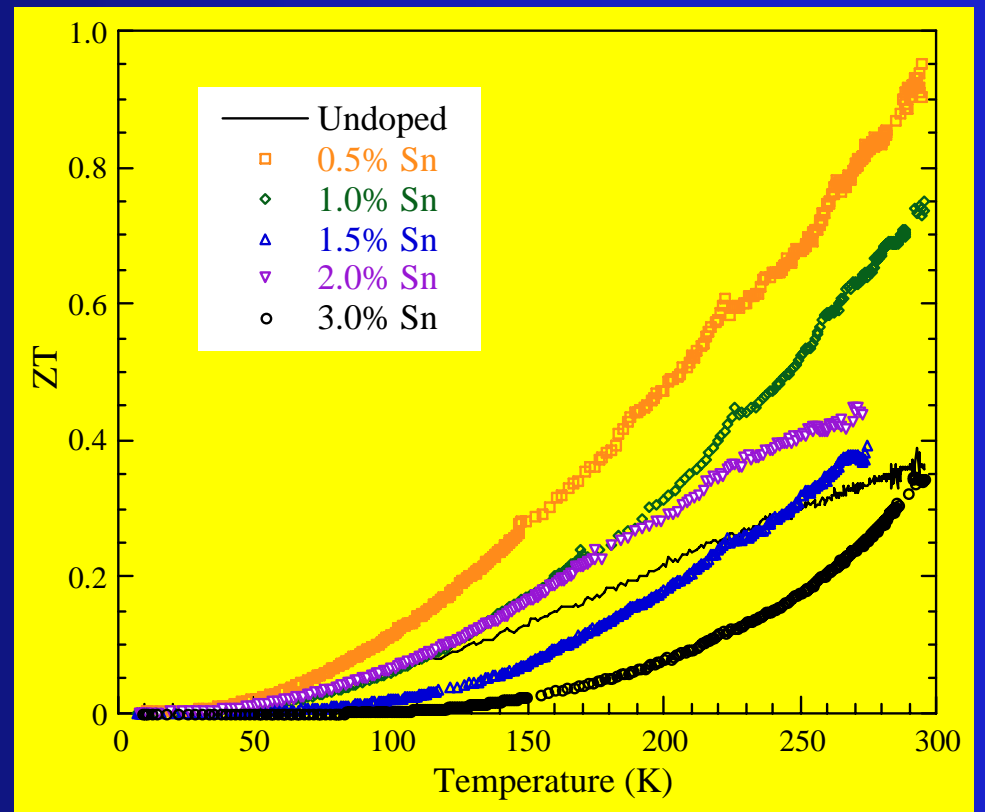
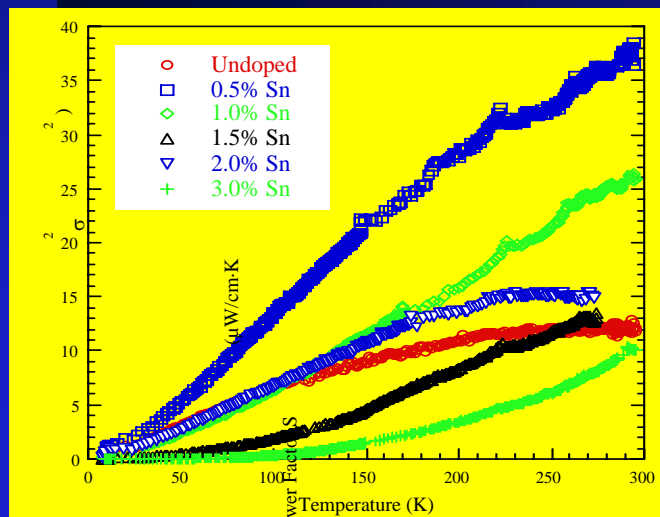
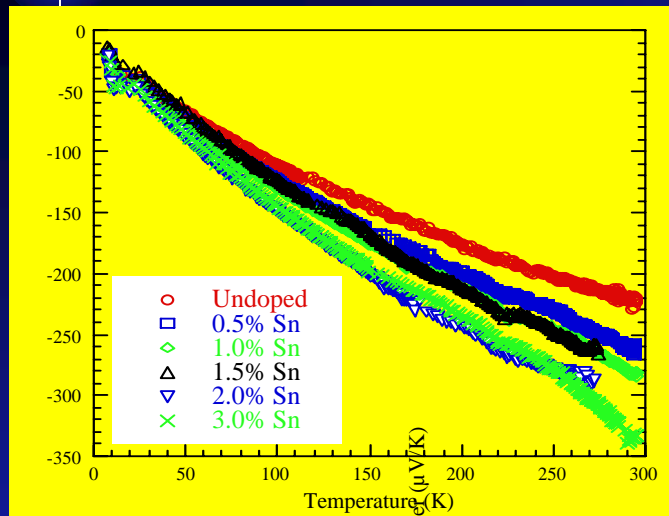




# $\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$ : Room temp $ZT=0.9$ . At 600 K estimated at 1.5 (to be verified)



# Sn Doping in $\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$



# Michigan State University /Tellurex Corp. Collaboration



New TE material grown at Tellurex



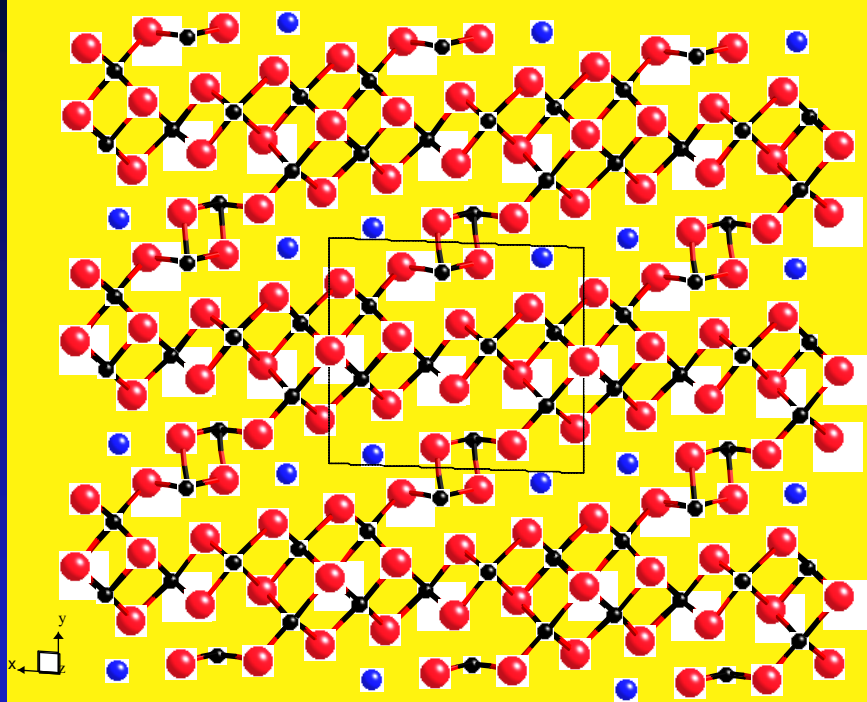
New TE material grown at MSU

Photo of the first TE module containing 63 couples  
 $n\text{-}\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}/p\text{-Bi}_2\text{Te}_3$

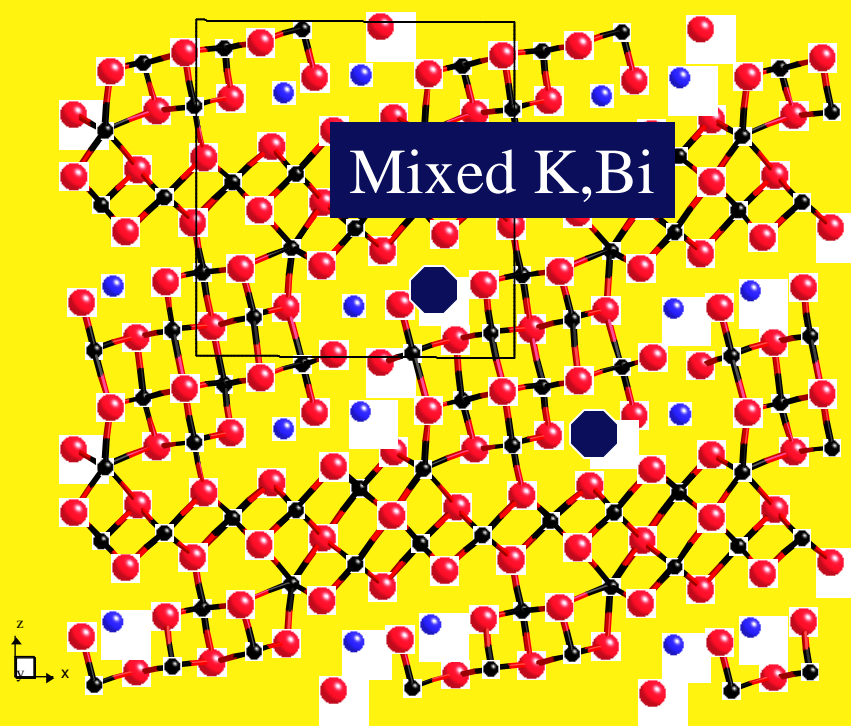


Unoptimized  
 $\Delta T = 36\text{ }^\circ\text{C}$   
 $T_h = 50\text{ }^\circ\text{C}$   
All materials grown  
at Tellurex Inc

# $\alpha\text{-K}_2\text{Bi}_8\text{Se}_{13}$ versus $\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$

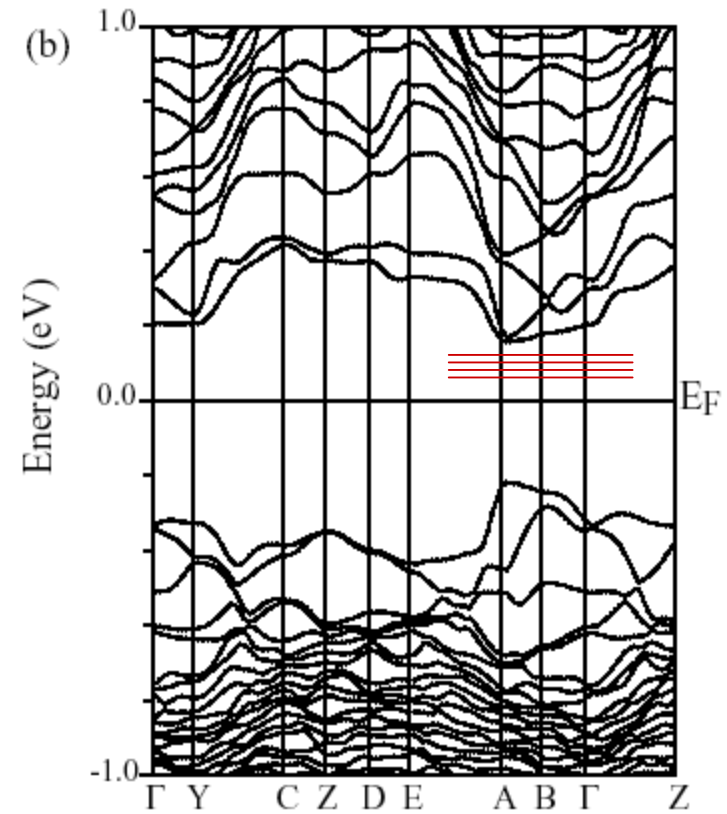
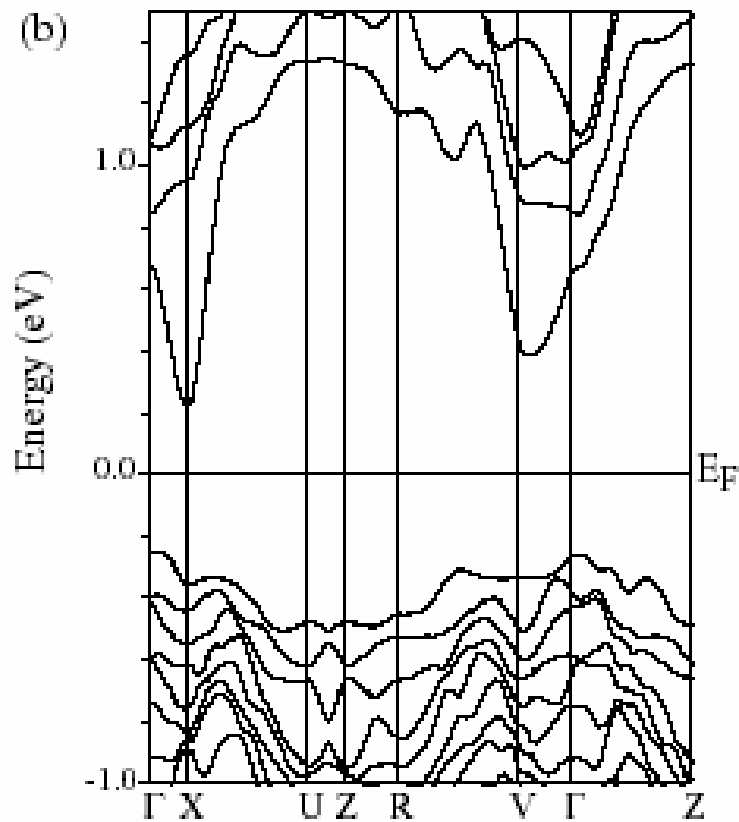


$\alpha\text{-K}_2\text{Bi}_8\text{Se}_{13}$ ,  $E_g=0.76$  eV



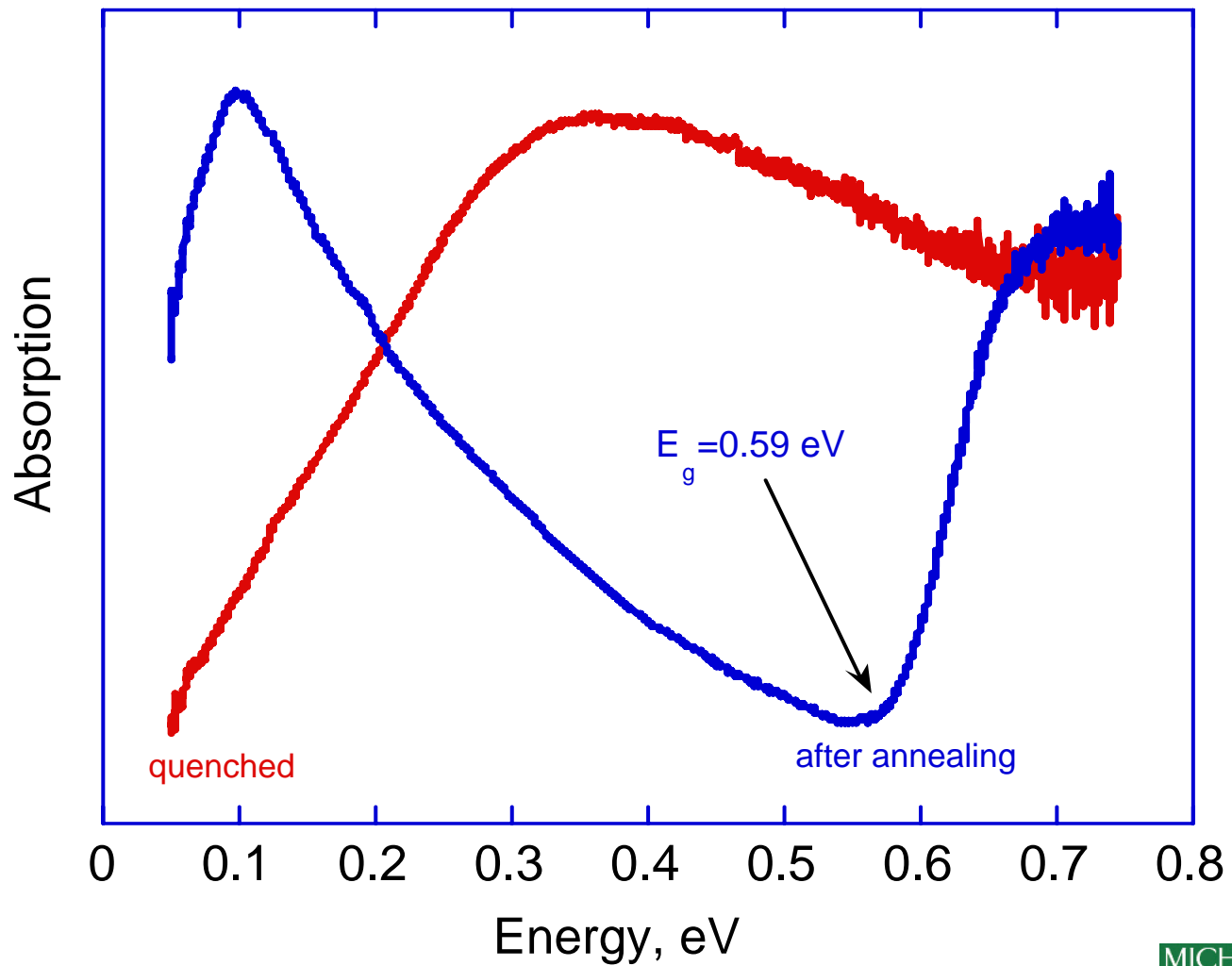
$\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$ ,  $E_g=0.59$  eV

# $\alpha$ -, $\beta$ - $\text{K}_2\text{Bi}_8\text{Se}_{13}$ : Electronic structure



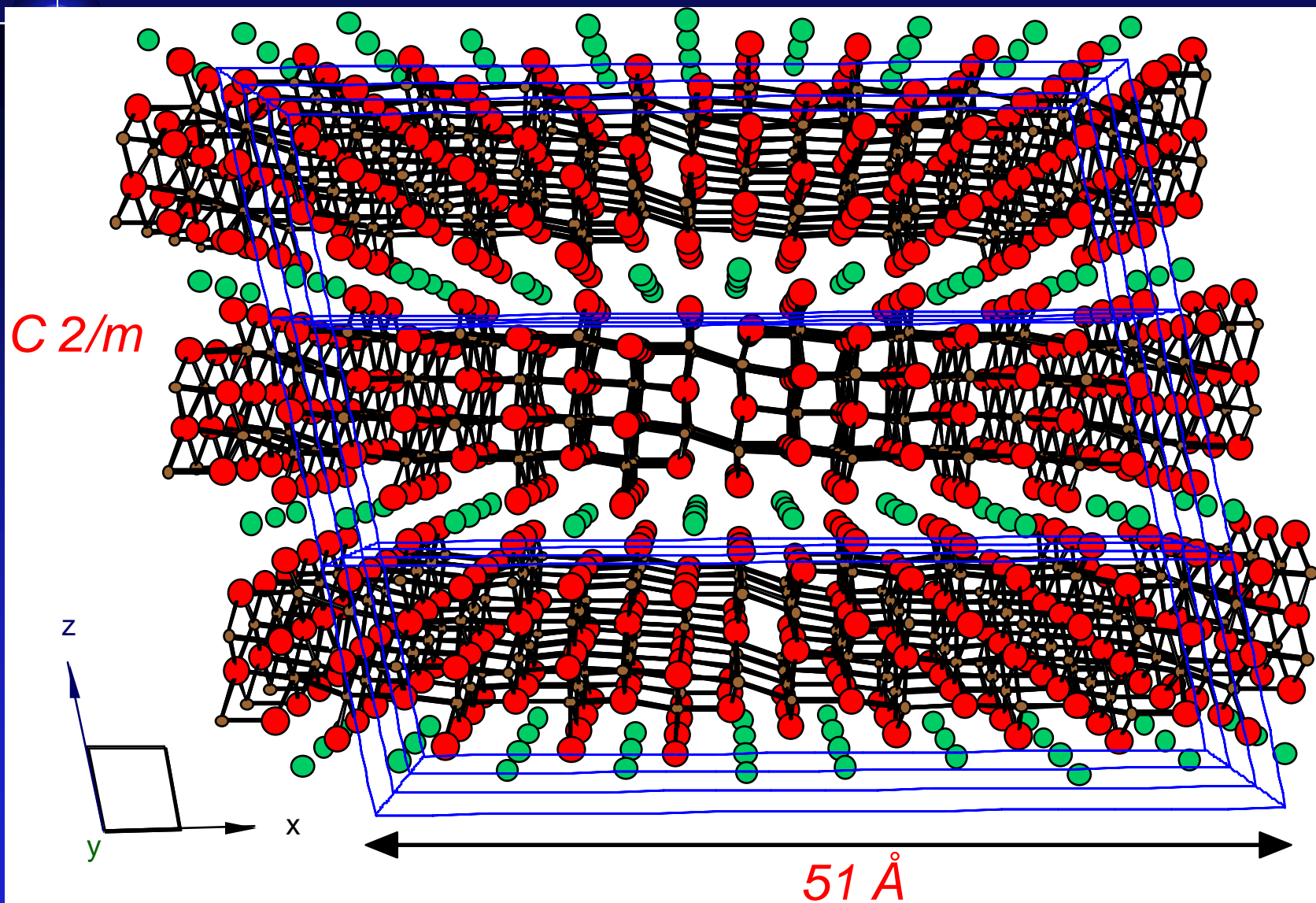
n-type character

# Quenched and annealed $\beta\text{-K}_2\text{Bi}_8\text{Se}_{13}$





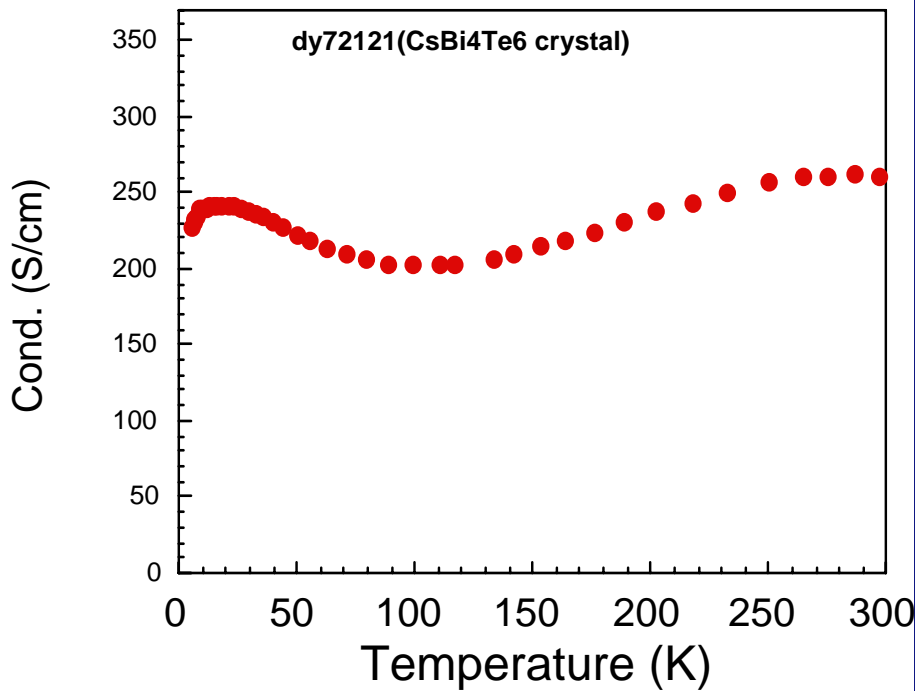
# CsBi<sub>4</sub>Te<sub>6</sub>



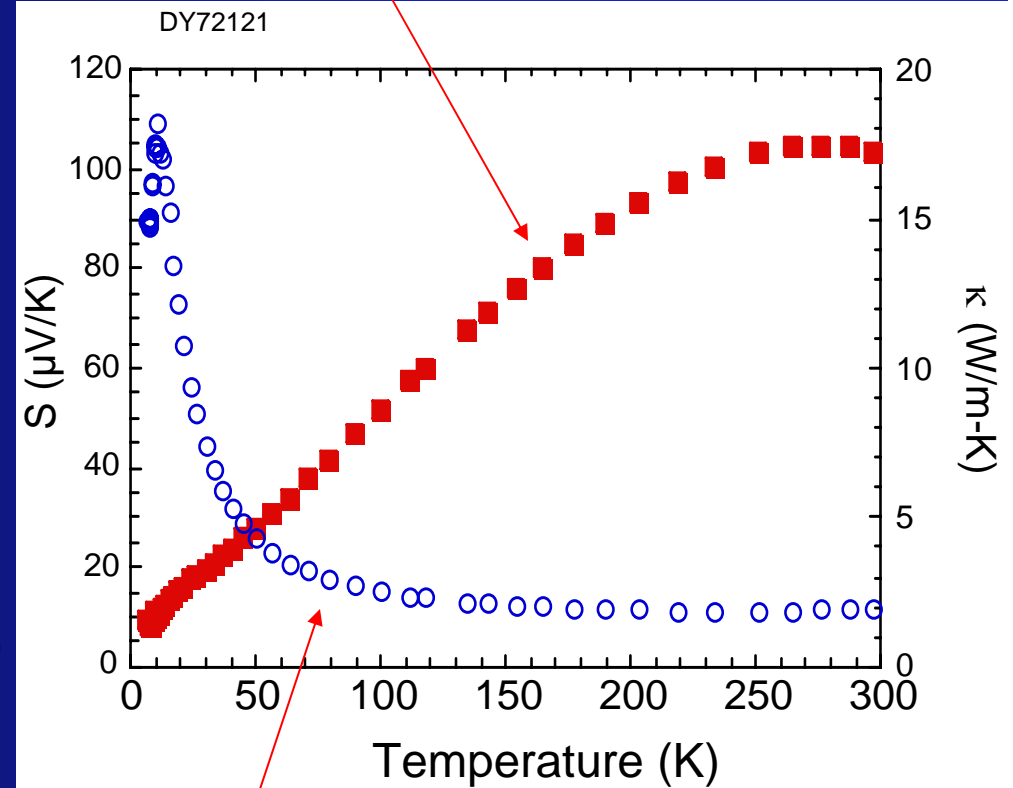


# “Undoped” as-prepared material

conductivity

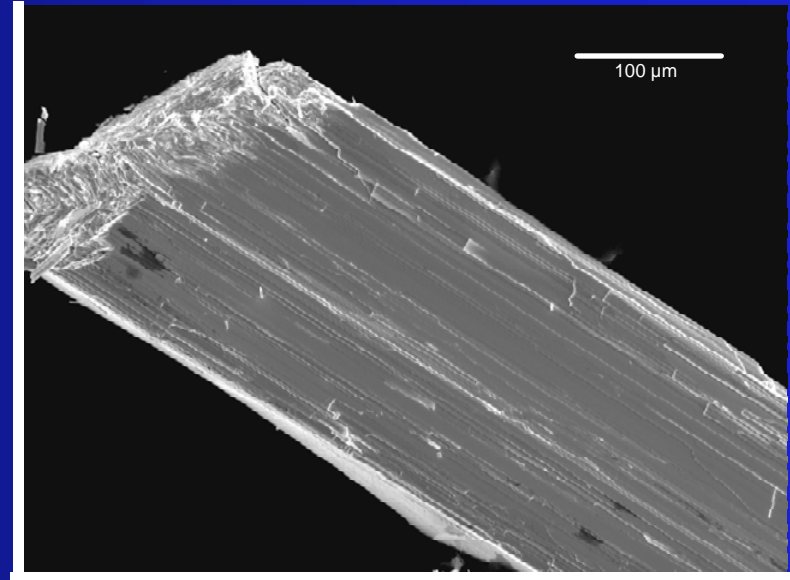
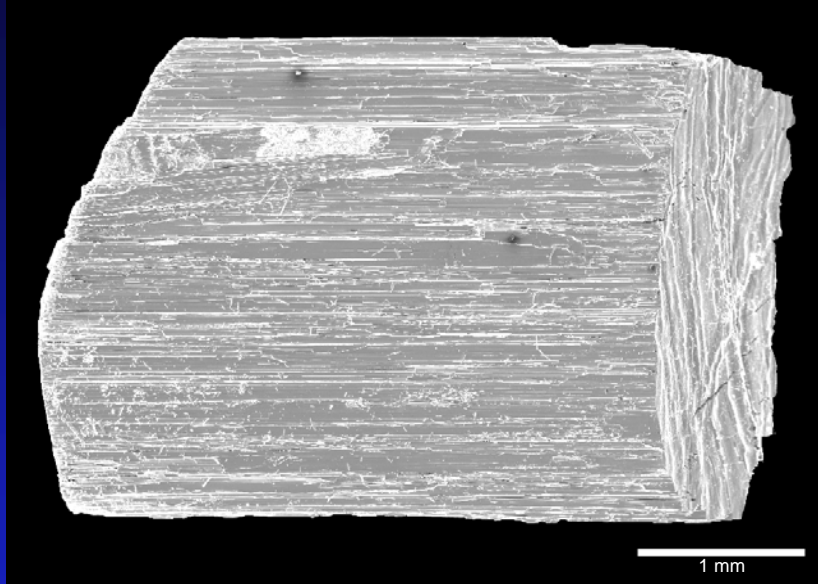


thermopower

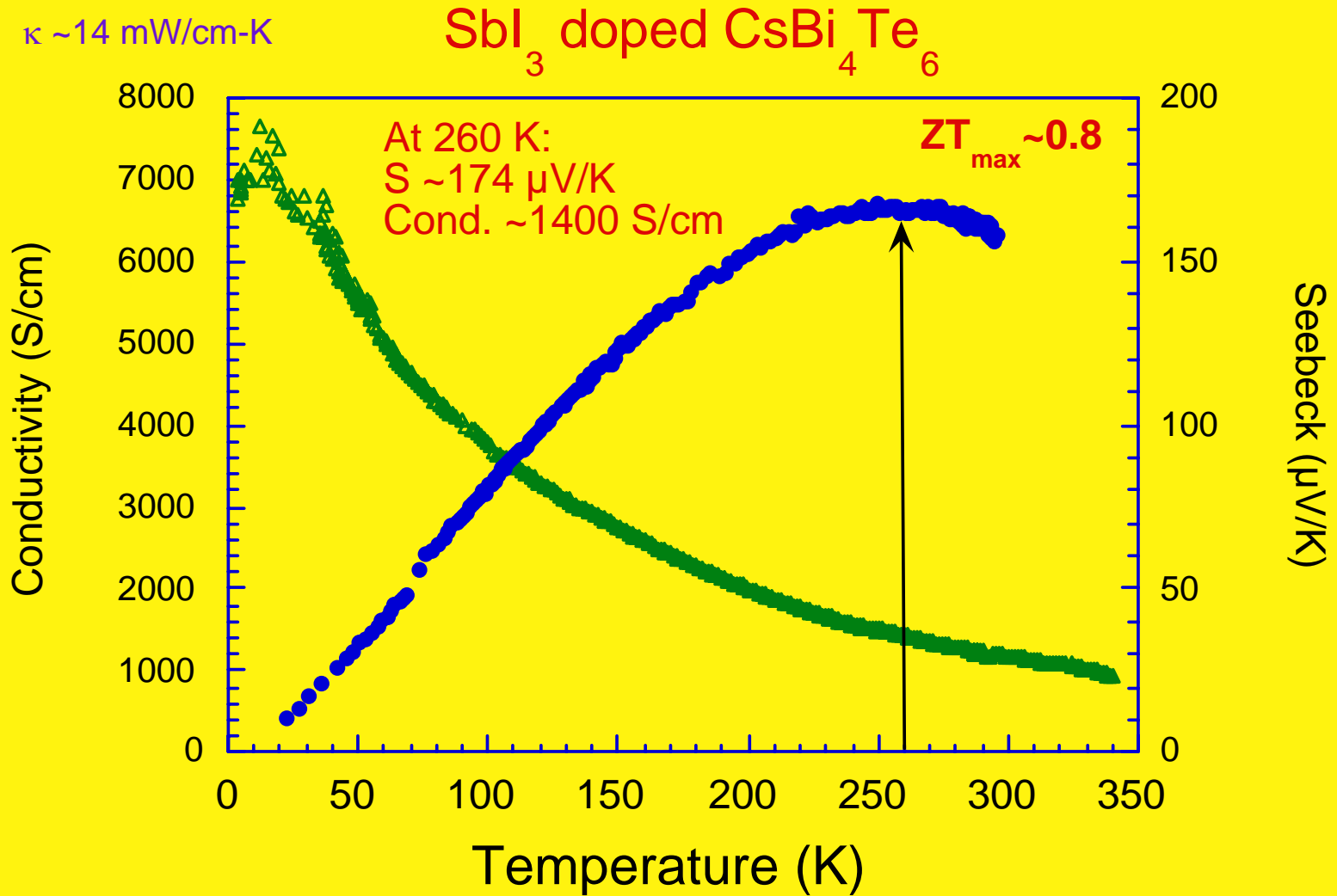


Thermal conductivity

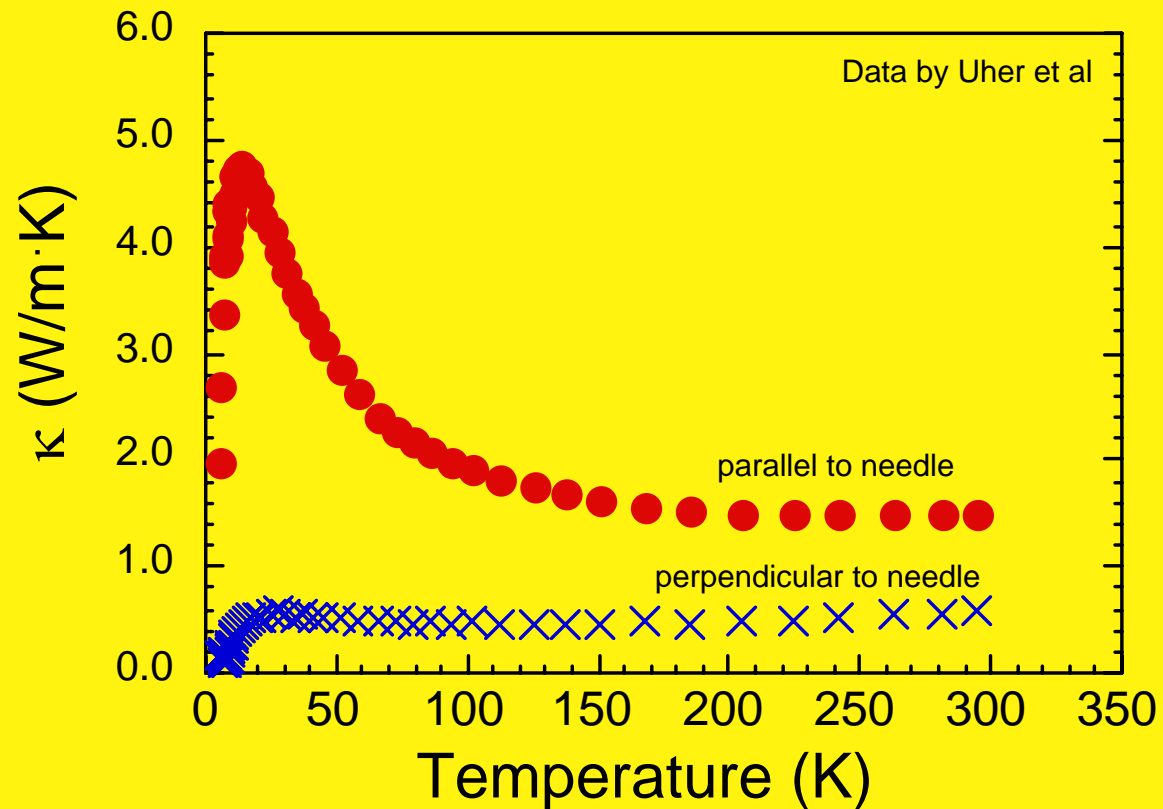
# Crystals of $\text{CsBi}_4\text{Te}_6$



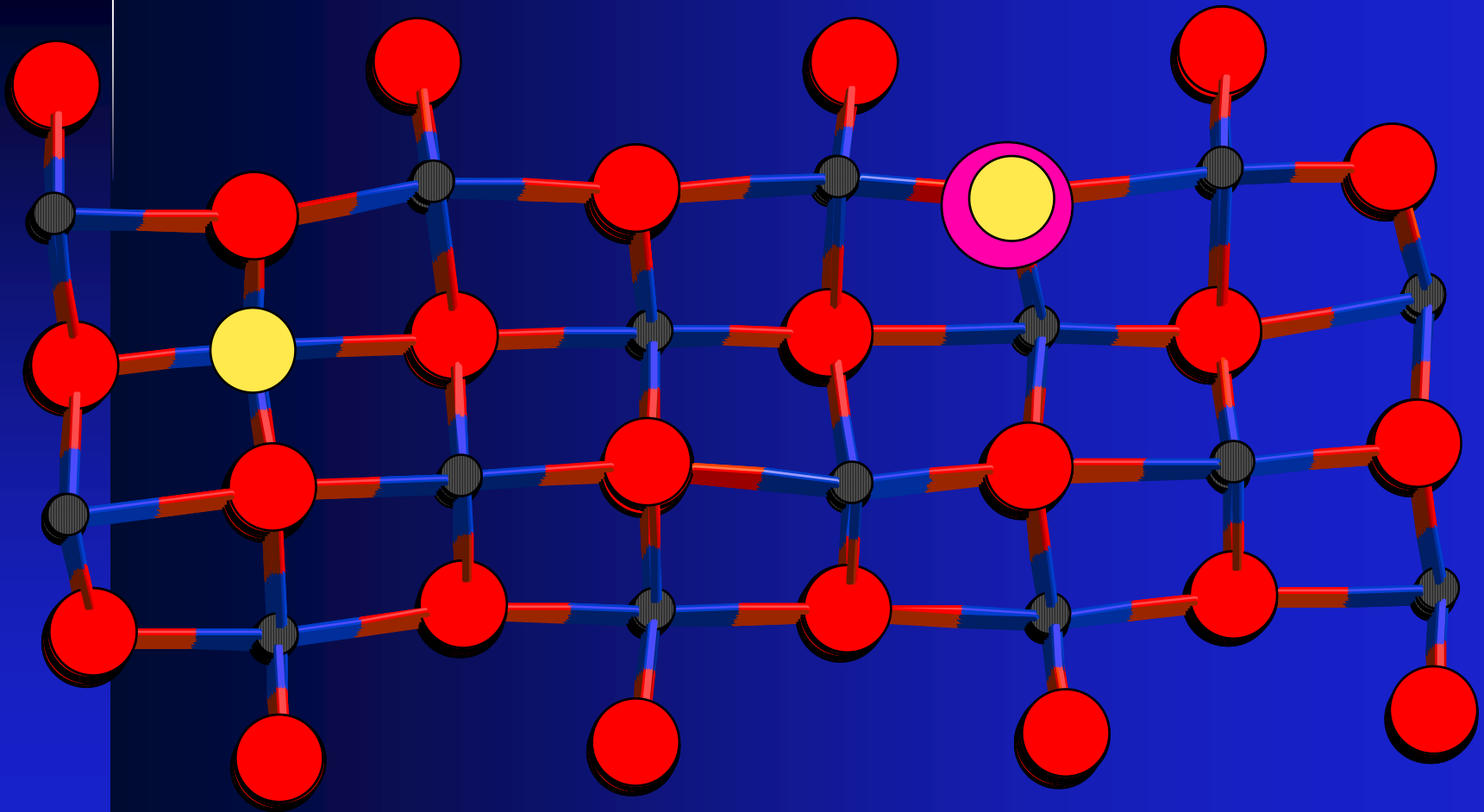
# Doped CsBi<sub>4</sub>Te<sub>6</sub>



# Thermal Conductivity of p-type $\text{CsBi}_4\text{Te}_6$

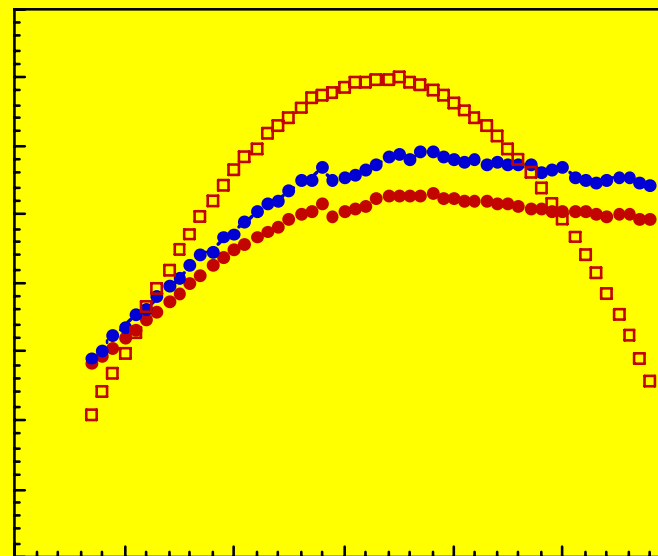
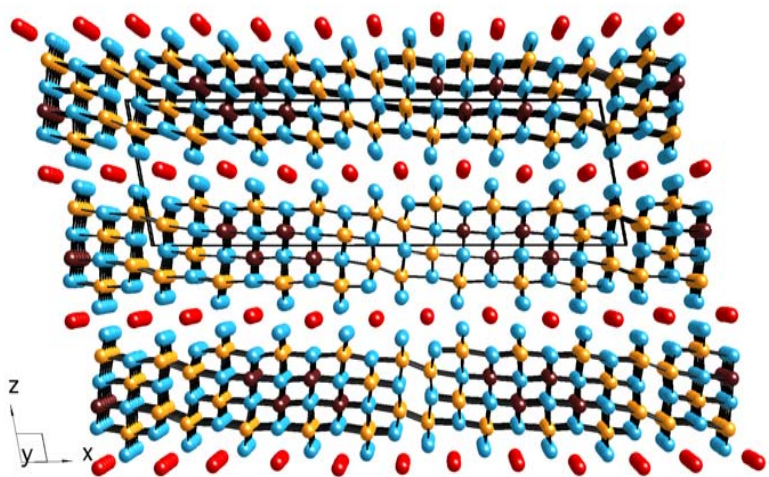


# Doping with $\text{SbI}_3$



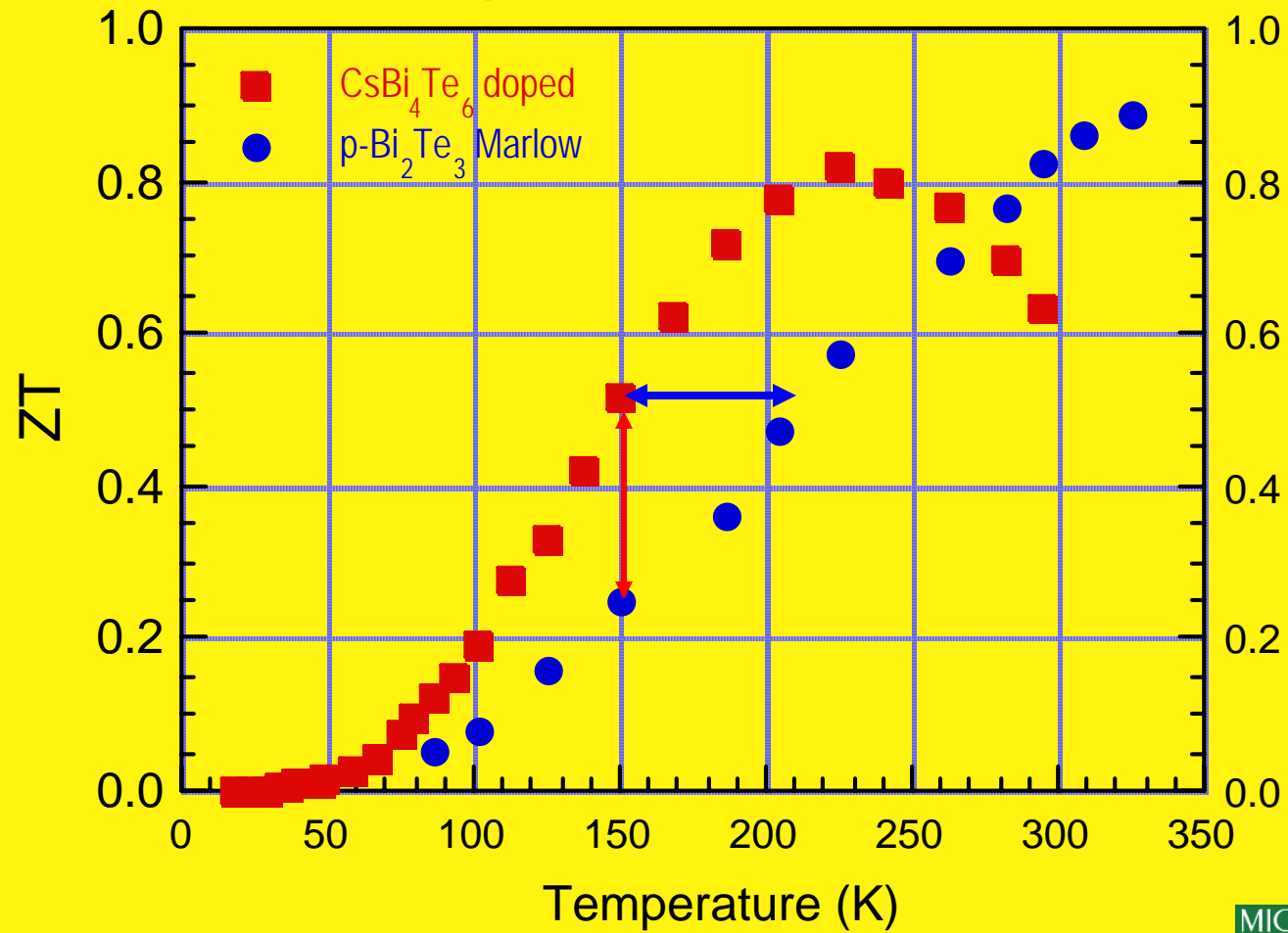


$x = 0.3$



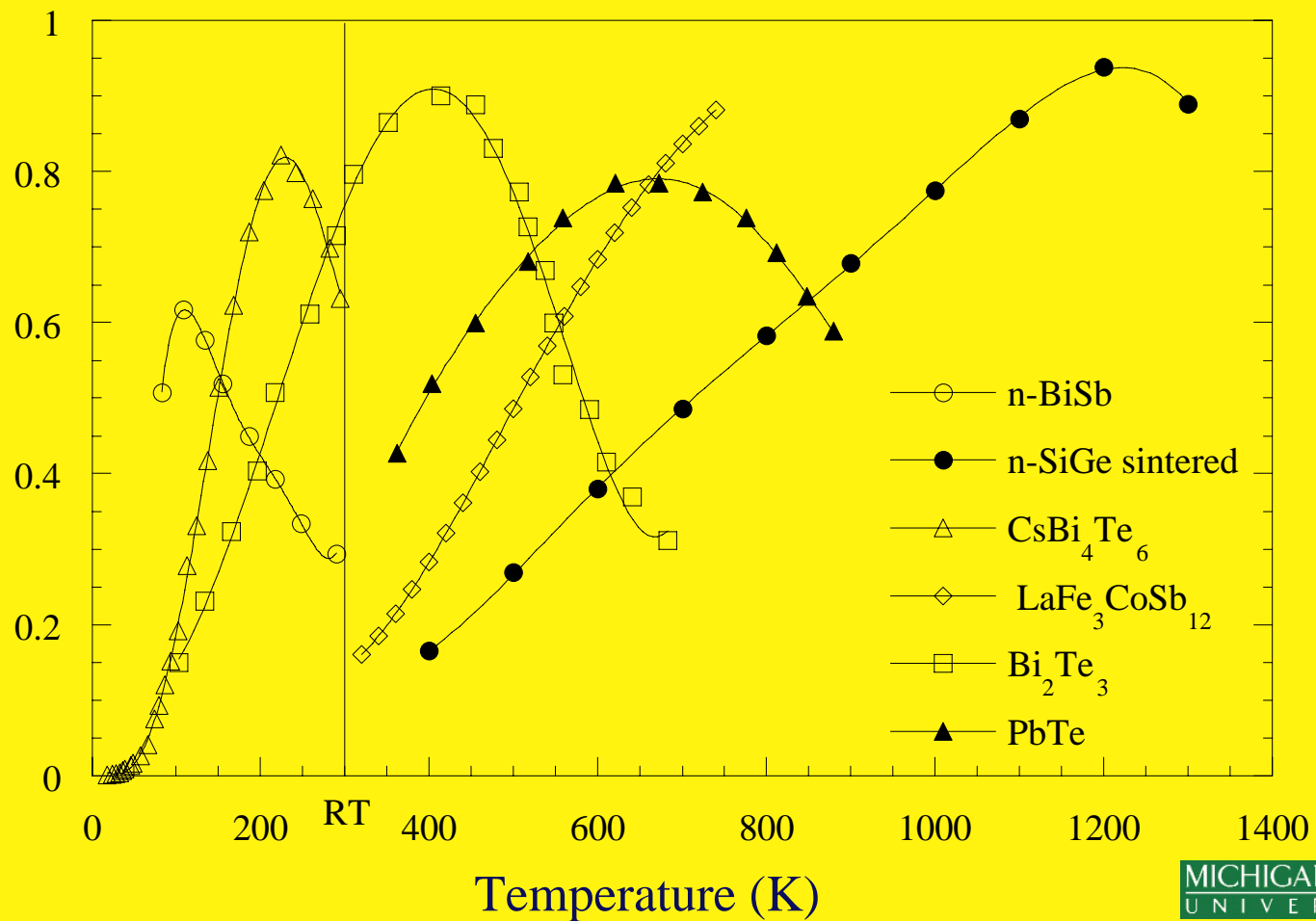
# CsBi<sub>4</sub>Te<sub>6</sub>

## Figure of Merit (ZT)



# Best TE Materials

ZT





# Conclusions

- The strategy to search for new materials in the  $(A_2Q)_n(PbQ)_m(Bi_2Q_3)_p$  (Q=Se, Te) system is successful
- Many new promising compounds identified
- All compounds strongly anisotropic
- Doping studies are important in ZT optimization
- ZT for  $\beta$ -K<sub>2</sub>Bi<sub>8</sub>Se<sub>13</sub> ~0.7 at rt, higher at >400K

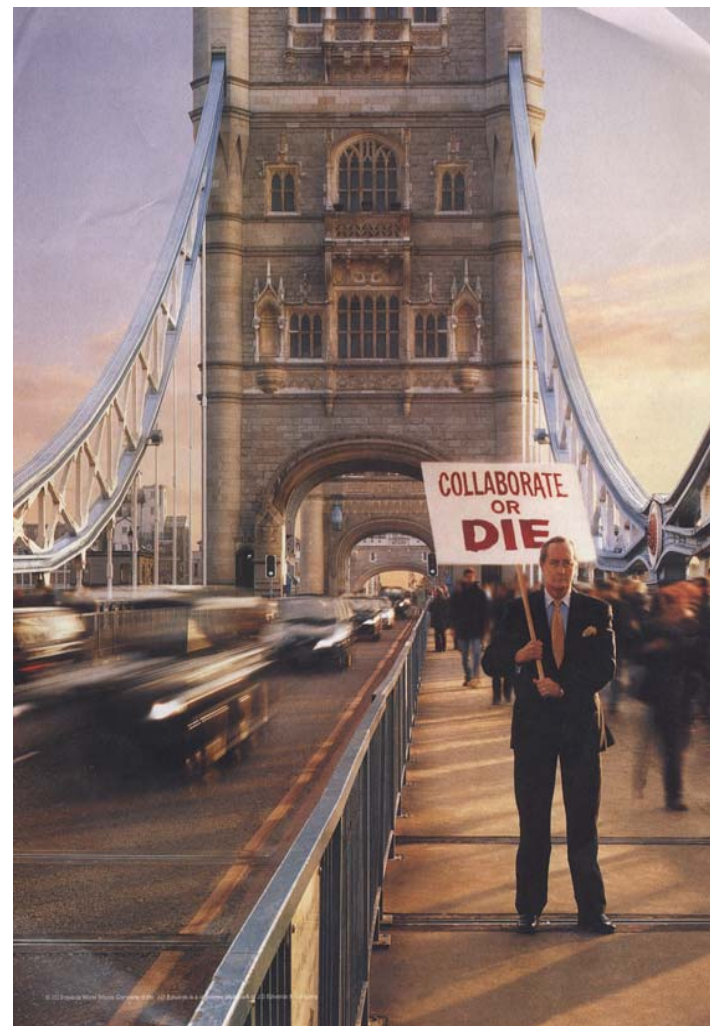
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(c) DiSalvo, F. J. “Thermoelectric cooling and power generation”, *Science*. **1999**, 285 5428, 703-706
- *Thermoelectric Materials 1998- The Next Generation Materials for Small-Scale Refrigeration and Power Generation Applications*, edited by Tritt, T. M.; Kanatzidis, M. G.; Mahan, G. D.; Lyon, Jr., H. B. *Mat. Res. Soc. Symp. Proc.* **1999**, Vol. 545, 233-246.

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